

# *Atmospheric electricity influencing biogeochemical processes in soils and sediments*

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Published Version

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Hunting, E. R., Harrison, R.G. ORCID: <https://orcid.org/0000-0003-0693-347X>, Bruder, A., van Bodegom, P. M., van der Geest, H. G., Kampfraath, A. A., Vorenhout, M., Admiraal, W., Cusell, C. and Gessner, M. O. (2019) Atmospheric electricity influencing biogeochemical processes in soils and sediments. *Frontiers in Physiology*, 10. 378. ISSN 1664-042X doi: <https://doi.org/10.3389/fphys.2019.00378> Available at <https://centaur.reading.ac.uk/83471/>

It is advisable to refer to the publisher's version if you intend to cite from the work. See [Guidance on citing](#).

To link to this article DOI: <http://dx.doi.org/10.3389/fphys.2019.00378>

Publisher: Frontiers

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# Atmospheric Electricity Influencing Biogeochemical Processes in Soils and Sediments

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## OPEN ACCESS

### Edited by:

Francesca Carella,  
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### Specialty section:

This article was submitted to  
Aquatic Physiology,  
a section of the journal  
Frontiers in Physiology

**Received:** 18 June 2018

**Accepted:** 19 March 2019

**Published:** 16 April 2019

### Citation:

Hunting ER, Harrison RG,  
Bruder A, van Bodegom PM,  
van der Geest HG, Kampfraath AA,  
Vorenhout M, Admiraal W, Cusell C  
and Gessner MO (2019) Atmospheric  
Electricity Influencing Biogeochemical  
Processes in Soils and Sediments.  
*Front. Physiol.* 10:378.  
doi: 10.3389/fphys.2019.00378

The Earth's subsurface represents a complex electrochemical environment that contains many electro-active chemical compounds that are relevant for a wide array of biologically driven ecosystem processes. Concentrations of many of these electro-active compounds within Earth's subsurface environments fluctuate during the day and over seasons. This has been observed for surface waters, sediments and continental soils. This variability can affect particularly small, relatively immobile organisms living in these environments. While various drivers have been identified, a comprehensive understanding of the causes and consequences of spatio-temporal variability in subsurface electrochemistry is still lacking. Here we propose that variations in atmospheric electricity (AE) can influence the electrochemical environments of soils, water bodies and their sediments, with implications that are likely relevant for a wide range of organisms and ecosystem processes. We tested this hypothesis in field and laboratory case studies. Based on measurements of subsurface redox conditions in soils and sediment, we found evidence for both local and global variation in AE with corresponding patterns in subsurface redox conditions. In the laboratory, bacterial respiratory responses, electron transport activity and H<sub>2</sub>S production were observed to be causally linked to changes in atmospheric cation concentrations. We argue that such patterns are part of an overlooked phenomenon. This recognition widens our conceptual understanding of chemical and biological processes in the Earth's subsurface and their interactions with the atmosphere and the physical environment.

**Keywords:** atmospheric electricity, bacterial respiration, biogeochemistry, Carnegie-curve, ions, redox potential

## INTRODUCTION

Concentrations of various chemical compounds in surface waters, soils and sediments have been observed to vary widely in both space and time, often detectable as diel (but also seasonal) fluctuations (e.g., Stockdale et al., 2009; Nimick et al., 2011; Smith et al., 2011). This variability is highly relevant for organisms that live within these spatio-temporally heterogeneous environments. While large mobile organisms interact at broader spatial scales, small and relatively immobile

organisms like bacteria, fungi or nematodes can be expected to be particularly sensitive to fluctuations in their immediate electrochemical environment. Small organisms respond to changes in their physico-chemical environment with changes in metabolic activity and behavior (e.g., Fenchel, 2002; Vanreusel et al., 2010). Disposal of respiratory electrons is essential for organisms to sustain metabolic activity that drives ecosystem processes, including respiration and the recycling of organic matter and nutrients (Cho and Azam, 1988). The availability of molecules accepting respiratory electrons (i.e., redox conditions) can hence pose an important constraint on the metabolic activity of organisms in soils and sediments (Hayes and Waldbauer, 2006).

While many studies have improved our understanding of processes governing the Earth's subsurface electrochemical environment, many observed variations remain difficult to reconcile with known drivers of electrochemical heterogeneity. Here, we briefly synthesize our understanding of drivers of Earth subsurface electrochemical variability, and present a novel conceptual foundation relating variations in atmospheric electricity (AE) to variations in the Earth's electrochemical environment and to consequences for the microorganisms living therein. We present evidence supporting the proposed linkages and identify challenges for future research.

## DRIVERS OF SPATIO-TEMPORAL VARIABILITY IN EARTH'S SUBSURFACE ELECTROCHEMISTRY

Small-scale variation in the electrochemical properties of sediments and soils are mainly controlled by biotic influences. For instance, locomotive activity of invertebrates that rework soils and sediments (bioturbation) is a well-known driver of micro to millimeter-scale redox conditions in both soils and sediments (Tokida et al., 2007; Hunting et al., 2012). Bacterial metabolic activity, in particular, is considered to be mainly controlled by this small-scale variation (Newman and Banfield, 2002). Redox fluctuations are likely an important selective pressure on microbes with repercussions for community composition and activity (Pett-Ridge and Firestone, 2005), for instance by selecting for metabolically more flexible bacterial taxa (DeAngelis et al., 2010). In turn, bacteria can secrete redox-active exudates (e.g., flavins) to maintain favorable redox conditions (Hunting and Kampfraath, 2013; Markelova et al., 2018), or can use long-distance (>1 cm) electron transfer to connect spatially separate bio-electrochemical processes (Nielsen et al., 2010; Pfeffer et al., 2012). Photosynthesis also promotes fluctuations in redox-conditions by introducing oxygen into the upper layers of soils and sediment (Battin et al., 2003; Laursen and Seitzinger, 2004), resulting in a net diurnal increase of oxygen concentrations and a net nocturnal decrease caused by respiration.

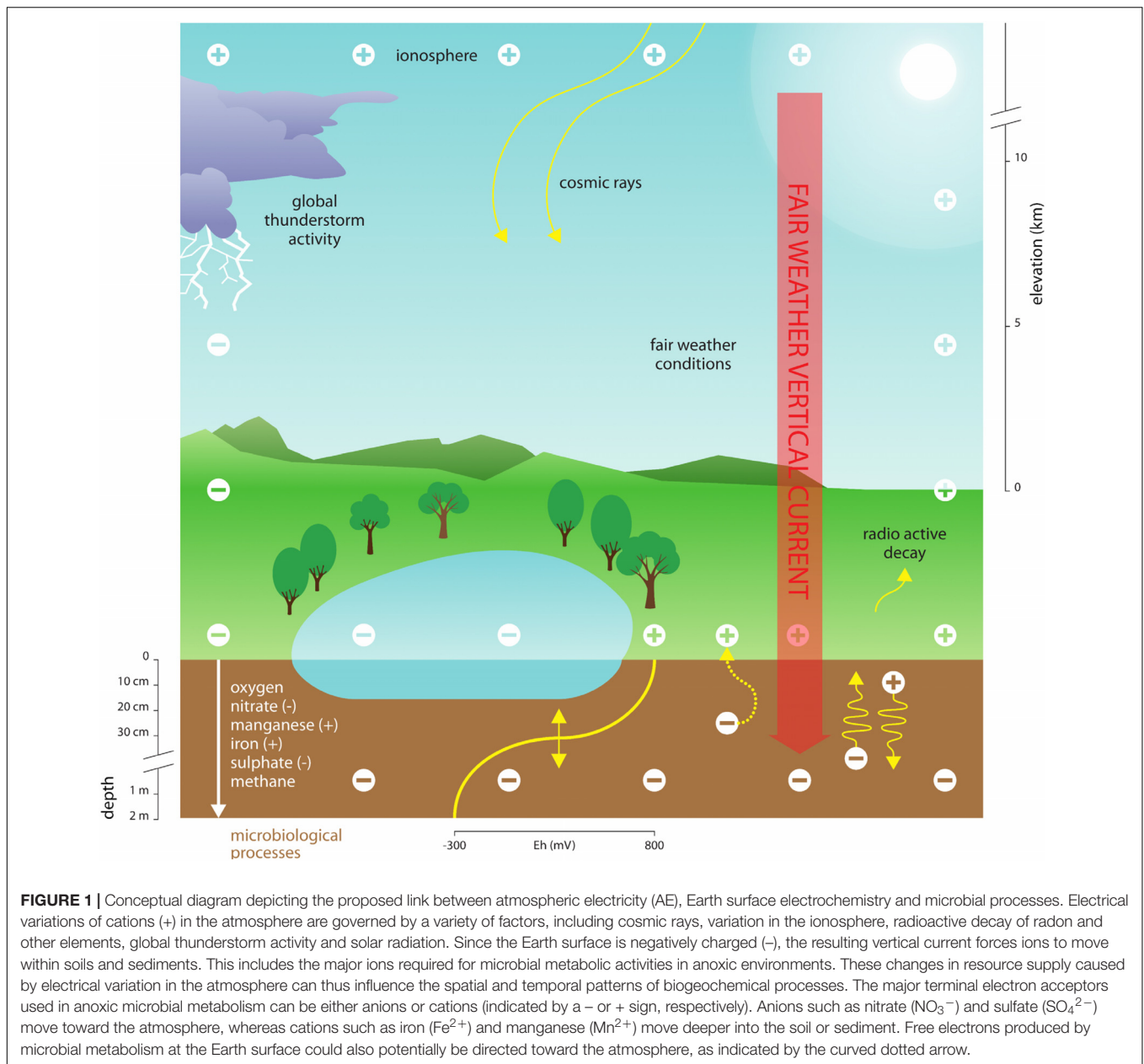
While small-scale variations are mainly driven by biological processes (Masscheleyn et al., 1991; Hayes and Waldbauer, 2006), diel and seasonal fluctuations of concentrations for many chemical species relevant to microbial processes (e.g.,

denitrification and methanogenesis) are also often linked over large distances (Lee, 1977; Laursen and Seitzinger, 2004; Allen et al., 2007; Spencer et al., 2007; Rusjan and Mikoš, 2010; Bass et al., 2013). The occurrence of large-scale temporal fluctuations in a wide variety of ecosystems suggests large-scale abiotic processes are also relevant to soil, sediment and water electrochemical properties (Scholefield et al., 2005). Indeed, various abiotic drivers of spatial linkages and synchronized temporal variability in subsurface chemical concentrations and microbial activity have been identified. They include solar activity, groundwater flow, atmospheric pressure, lunisolar and tidal cycles, and gradients of the chemical potential of charge carriers (reviewed in Lanzerotti and Gregori, 1986; Tokida et al., 2007). In inland waters and terrestrial soils, charge separation in clay or other minerals, contaminants and ground-water flow have also been shown to influence the electrochemical environment (e.g., Revil and Jardani, 2013).

Despite the breadth of understanding of processes governing the Earth's subsurface electrochemical environment and the consequences for organisms, the known drivers fail to explain all observed electrochemical variations. This is especially true for variations in the deeper layers (up to meters) of Earth's surface (Vorenhout et al., 2011). For instance, while photosynthesis can be responsible for diel variation of redox-conditions in biofilms and surficial (<1 cm) soil and sediment layers (Battin et al., 2003; Laursen and Seitzinger, 2004), it unlikely affects deeper environments and associated organisms, since oxygen diffusion is slow and consumption by heterotrophs is fast (Laursen and Seitzinger, 2004). Here, we propose a new perspective based on the idea that variation in AE is an additional factor underlying cyclic variation in the electrochemistry and associated microbial communities and activities in the Earth's subsurface environment.

## CONCEPTUAL FOUNDATION OF RELATIONSHIPS BETWEEN ATMOSPHERIC ELECTRICITY, EARTH'S SUBSURFACE ELECTROCHEMISTRY AND MICROBIAL COMMUNITIES

Electrical properties of the near-surface atmosphere (e.g., ion concentrations and the atmospheric potential gradient) vary on daily and seasonal times scales (Israelsson and Tammet, 2001; Harrison, 2004). An atmospheric electric field is present even in fair-weather regions as a consequence of global electric current flows driven by thunderstorm regions (e.g., Rycroft et al., 2000; Haldoupis et al., 2017). Locally, environmental conditions, radioactive decay of radon, charges of aerosols, and atmospheric pollution may further contribute to variation in atmospheric electric conditions (Matthews et al., 2019). The combination of global and local variations in AE leads to variations over various spatial and temporal scales, with diel fluctuations being particularly important (Israelsson and Tammet, 2001). The universal diel pattern in vertical current and potential gradient is dominated by a minimum at around 04 Universal Time



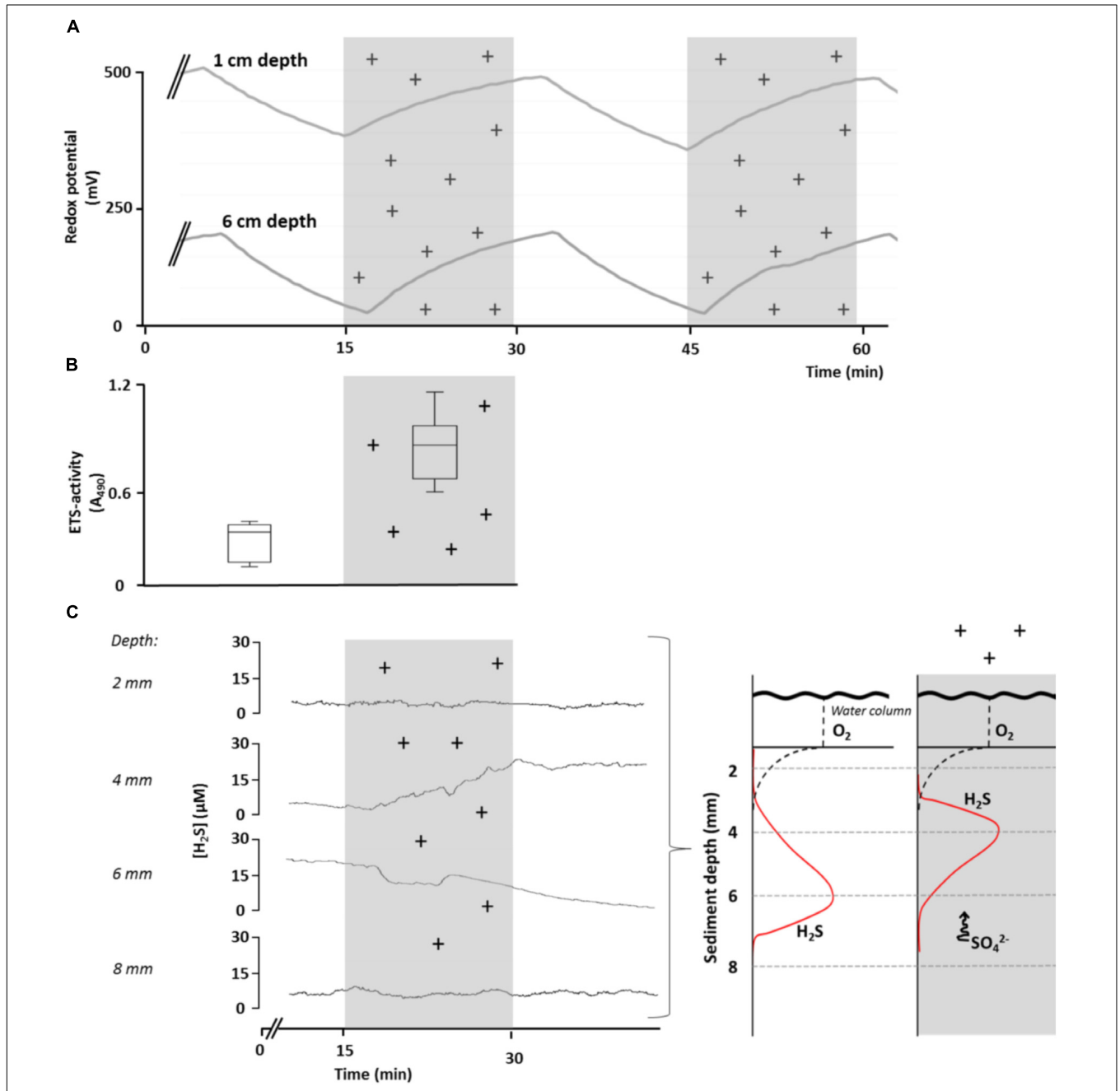
(UT) and a maximum at around 19 UT. This universal pattern is clearest in clean maritime air where aerosol pollution and other local sources of variation (e.g., diel fluctuations in radon concentration) are minimized. In contrast, over land diel patterns are more influenced by local variations in AE (Israelsson and Tammet, 2001; Harrison, 2004).

Electric currents rely on movement of (small) ions in the atmospheric electric field, and typically range between 0.5 and  $3.0 \text{ pA m}^{-2}$  at the Earth surface interface, where the currents subsequently enter the Earth surface as part of the global electric circuit (Rycroft et al., 2008; Harrison, 2013). Other geophysical processes (e.g., groundwater flow) contribute to influencing the electrical properties of the Earth subsurface (Lanzerotti and Gregori, 1986; Wada and Umegaki, 2001; Revil

et al., 2010). In soils, water bodies, and their sediments, currents induced by variations in AE likely influence the release of respiratory electrons and movement of ions, thereby critically affecting redox conditions with repercussions particularly for microorganisms. For instance, variations in AE could induce the vertical movement of charged terminal electron acceptors that are essential for microbial respiration (see **Figure 1** for a conceptual diagram). Terminal electron acceptors relevant to microorganisms (e.g.,  $\text{NO}_3^-$ ,  $\text{Mn}_4^{3+}$ , and  $\text{SO}_4^{2-}$ ) differ in size and charge, suggesting that they move at different speeds within Earth subsurface environments. Moreover, ion movement is influenced by the electrical conductivity of water, soils and sediments. Surface soil layers, for instance, typically have a conductivity of 0.1–2.0 dS/m (Rhoades and Corwin, 1981), which

is low compared to seawater, for instance ( $\sim 4$  S/m; e.g., Al-Shamma'a et al., 2004), and may impede long-distance (Revil et al., 2010) but not short-distance ( $\mu\text{m} - \text{mm} - \text{cm}$ ) ion movement (Wada and Umegaki, 2001; Mann et al., 2005). Such variation in redox properties of soils and sediments driven by

AE likely affects the ability of microbes to dispose of their respiratory electrons (Figure 1). To date, the consequences of variation in AE on the electrochemical properties of subsurface ecosystems and the organisms living therein remains entirely unexplored.



**FIGURE 2 |** Effect of elevated levels of atmospheric cations on redox conditions, bacterial respiratory activity and  $\text{H}_2\text{S}$  concentrations (sulfate reduction) in sediments of aquatic microcosms. **(A)** Redox potential,  $E_h$ , was measured at 1 and 6 cm sediment depth in response to manipulated atmospheric cation concentrations. **(B)** Bacterial respiratory activity, measured as electron transport system (ETS) activity (expressed as relative absorption at 490 nm), was significantly lower in control microcosms than in microcosms in which the atmosphere was ionized for 24 h ( $t$ -test,  $p = 0.002$ ,  $n = 6$ ). **(C)** Changes in bacterial  $\text{H}_2\text{S}$  concentration at different sediment depths in response to ionization of the overlying atmosphere (left panel), suggesting a shift in  $\text{H}_2\text{S}$  production toward the surface as a result of upward  $\text{SO}_4^{2-}$  movement (conceptually depicted in right panel). Shaded areas and (+) indicate periods of experimental ionization. A single time-series measurements is presented for clarity and considered representative of replicate ( $n = 12$ ) runs.



## EVIDENCE FOR LINKAGES BETWEEN ATMOSPHERIC ELECTRICITY, SUBSURFACE ELECTROCHEMISTRY AND MICROBIAL METABOLIC ACTIVITY

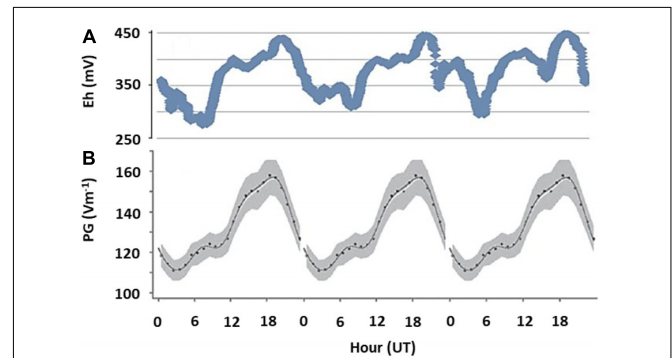
### Laboratory Experiments

We conducted several laboratory experiments to explore effects of variation in AE on sediment redox conditions and bacterial metabolism (for a description of the experimental approach see **Supplementary Material S1**). We found that sediment redox potential ( $E_h$ ) in aquatic microcosms evolved independently of sediment pH (max. change  $\pm 0.1$  units) or oxygen concentration (max. change  $\pm 1\%$  saturation), when exposing them to experimentally manipulated levels of atmospheric ion concentrations. In contrast,  $E_h$  gradually increased at different sediment depths, starting immediately when ionization began, then declined and quickly stabilized when disrupting ionization (**Figure 2A**). Control microcosms in which the overlying atmosphere was not ionized soon reached a redox equilibrium that remained constant throughout the experiment (data not shown). No effects on  $E_h$  in the sediment were found after exposing the microcosms to radiation [UV A, B, and C as well as photosynthetically active radiation (PAR) and infrared; data not shown]. Taken together, our empirical data shows that fluctuations of  $E_h$  in the microcosm sediment were independent of solar radiation but were strongly influenced by the manipulated shifts in ion concentrations in the overlying atmosphere. These findings provide clear evidence that variations in AE have potential to influence geochemical and microbial processes via alterations of  $E_h$ .

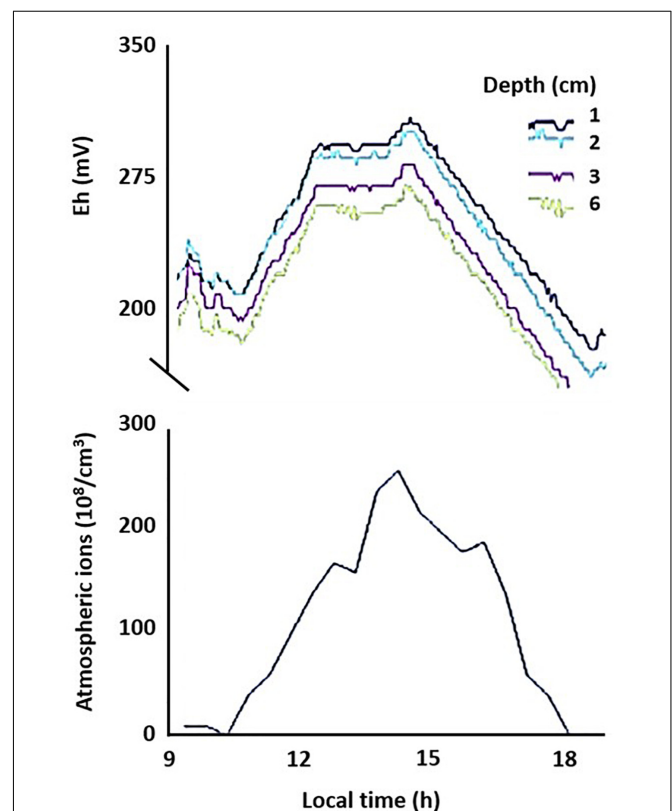
We then tested the response of bacterial communities in sediments of microcosms to fluctuations in  $E_h$  induced by variations in AE laboratory conditions by increasing concentrations of ions in the overlying atmosphere. We measured the respiration of the bacterial community as ETS activity in the upper sediment layer (<1 cm) after 1 day of exposure to atmospheric ionization and observed a two-fold increase (*t*-test,  $p < 0.05$ ) compared to control microcosms (**Figure 2B**). However, since multiple bacterial processes can contribute to ETS activity, the cause of the increase remains uncertain.

To examine the issue further, we experimentally exposed aquatic microcosms to increased concentrations of ions in the overlying atmosphere and assessed  $H_2S$  concentrations in response to atmospheric ionization. We chose  $H_2S$  concentration as the most informative response variable, since it results directly from  $SO_4^{2-}$  reduction. Moreover, since  $H_2S$  does not carry a charge, any changes in  $H_2S$  due to variation in AE can only result from changes in  $SO_4^{2-}$  reduction. We observed a gradual increase in  $H_2S$  concentrations at 4 mm below the sediment surface, whereas  $H_2S$  concentrations decreased at 6 mm below the sediment surface. The response in  $H_2S$  concentrations after the start of ionization was slightly delayed (**Figure 2C**). These findings suggest that the depth of maximum  $SO_4^{2-}$  concentrations shifted toward the sediment surface in response to ionization

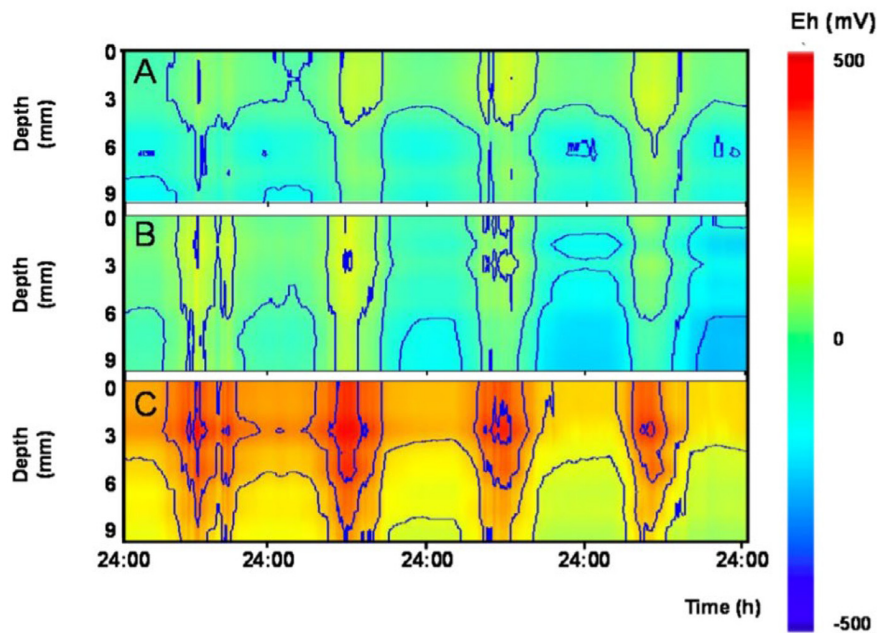
where the microbial community quickly responded by reducing  $SO_4^{2-}$  to  $H_2S$ .



**FIGURE 3** | Fair weather diel fluctuations in sediment redox potential and AE. **(A)** Redox potential ( $E_h$ ) measured at 10 cm depth of a natural pristine sediment as a single run over 3 days in Lake Cadagno, an alpine lake in Switzerland, in October 2017. **(B)** The universal periodicity in atmospheric electrical properties (expressed as potential gradient, PG, between atmosphere and ground), which is visible around the globe during fair weather conditions (Harrison, 2013). All data is plotted against universal time, UT, and represent single time-series measurements. Peaks at 19 UT indicate that fluctuations in redox conditions are governed by global patterns in AE.



**FIGURE 4** | Coherence of temporal changes in net atmospheric cation concentrations and sediment redox conditions. Redox potential ( $E_h$ ) was measured on 24 October 2013 at different depths (1, 2, 3, and 6 cm) in sandy sediment of a ditch in Netherlands. Data of this single run is plotted against local time (GMT +2).



**FIGURE 5** | Fair weather diel rhythms in sediment redox potential outdoor mesocosms experiencing different levels of bioturbation. Contour plots show depth profiles (0–9 mm) redox potential ( $E_h$ ) over 4 days in three different mesocosms. Mesocosms contained different combinations of invertebrates known to rework surface sediment to various degrees: **(A)** no bioturbation: invertebrates lacking; **(B)** low level of bioturbation: *Tubifex* spp. and *Asellus aquaticus*; and **(C)** high level of bioturbation: *Gammarus pulex*, *Asellus aquaticus*, *Chironomus riparius*, *Tubifex* spp., and *Lumbriculus variegatus*.

## Field Observations

To evaluate whether the links between AE and subsurface redox variations observed in microcosm experiments apply in realistic settings, we measured  $E_h$  in (1) an outdoor mesocosms facility containing no or different combinations of invertebrates causing bioturbation of surface sediments; (2) freshwater sediments at two distinct sites, and (3) soils at geographically distinct locations (see **Supplementary Material S1** for details). In the shallow littoral zone of pristine Lake Cadagno in the Swiss Alps, we observed diel fluctuations in  $E_h$  following the universal cycle in atmospheric potential gradient, with peaks occurring at around 19 UT (**Figure 3**). In contrast, in a ditch experiencing urban pressure in Netherlands, the diel fluctuation was dominated by local influences, with a peak occurring at around 2–4 pm local time (**Figure 4**). However, the number of cations in the ground-level atmosphere also appeared to covary with sediment  $E_h$  (**Figure 4**).

Diel fluctuations in  $E_h$  in response to local variation in AE were also observed in sediments of freshwater outdoor mesocosms in Netherlands (**Figure 5** and **Supplementary Figure S1**). Here, diel patterns were visible even in the presence of invertebrates reworking the upper sediment layers, and were more pronounced during fair-weather conditions than on cloudy days (**Supplementary Figure S1**). These diel rhythms were not observed to covary with other tested meteorological variables such as solar radiation, temperature and air pressure (data not shown). This finding indicates that the postulated link between AE and sediment redox potential persists also when major hydrological and geophysical processes (e.g., groundwater flow) are excluded. This supports our hypothesis that a direct link exists

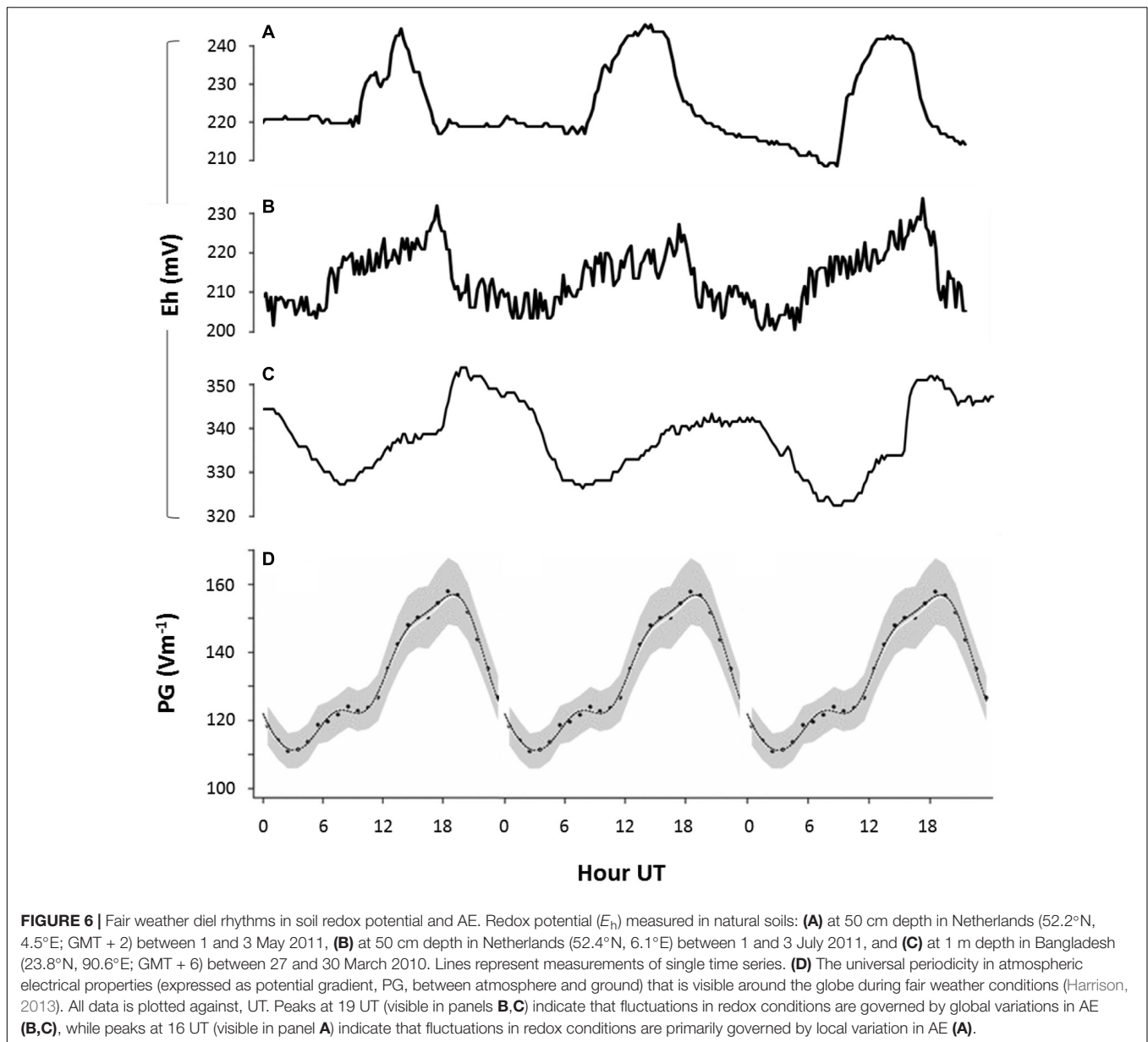
between subsurface  $E_h$  and AE. Interestingly, natural freezing of the top water layer in the mesocosms served as an unplanned experimental control, since the co variation between subsurface  $E_h$  and AE disappeared, probably as a consequence of poor conductive properties of ice (data not shown). Finally,  $E_h$  in soils at three distant sites also followed diel patterns in AE whose influence extended relatively deep into the soil (typically 50–100 cm; **Figure 6**).

These outdoor measurements suggest that variation in subsurface redox potential can follow both the universal diel cycle in the atmospheric potential gradient (**Figure 6**) and local sources of variation affecting atmospheric cation concentrations at ground level (<1 m). Together with our laboratory results, these field observations reveal that both global and local variations in AE influence redox conditions and microbial processes in soils and sediments, in which strong local influences on redox patterns can dominate in some locations.

## IMPLICATIONS AND PERSPECTIVES

The findings from our laboratory experiments and field observations support the hypothesis that variation in AE can influence  $E_h$  in different soil and sediment matrices with consequences for microbial communities in these environments. The significance of this phenomenon in natural settings remains unclear, however, because persisting knowledge gaps impede a conclusive understanding of the causal relationships between AE and Earth's electrical environment. Challenges for future research range from elucidating the relevant scales of the physical and





chemical linkages to how these links directly or indirectly govern distinct groups of organisms.

How specific chemical species and organisms respond to changes in AE likely depends on the relative magnitude of a combination of various physical sources of variations. These include solar activity, groundwater flow, gradients of the chemical potential of charge carriers (reviewed in Lanzerotti and Gregori, 1986; Revil and Jardani, 2013), as well as electrochemical soil and sediment properties, including electrical resistance and the size and charge of terminal electron acceptors. Unraveling the absolute and relative roles of regional and global-scale drivers of variation in AE (März and Harrison, 2003; Harrison, 2004) and redox potentials in water, sediments and soils would thus be a promising, though challenging research field. Earth subsurface electrochemistry varies in depth, and hence future work needs

to assess to which extent variations in AE can translate to synchronized responses in Earth subsurface electrochemistry in relation to the conditions at different soil and sediment depths (e.g., moisture, conductivity,  $E_h$ ). Likewise, it is necessary to partition the role of AE in relation to other major drivers of Earth subsurface electrochemistry. This requires coordinated field experiments over a wide range of geographical locations to assess the significance on a global scale and the importance of local influences in superimposing universal rhythms.

The coupling between AE and subsurface electrochemistry observed in our field and laboratory studies also suggests that microorganisms in these environments are vulnerable to anthropogenic influences affecting variation in AE. In particular, anthropogenic pollution by smoke, sulfur dioxide and aerosols can affect AE (Retalis et al., 1991; Sheftel et al., 1994;

Kamra and Deshpande, 1995). Our results suggest that such pollutants could have strong, though currently unknown, indirect effects on subsurface microorganisms and processes by affecting temporal patterns of AE. Furthermore, electrical pollution by high-voltage power lines (mostly operated with alternating currents) is a common local factor affecting variations in AE (Maruvada, 2012). The resulting static electric fields have been observed to trigger responses in a wide array of organisms, particularly behavioral responses in invertebrates (Petri et al., 2017; Schmiedchen et al., 2018). However, these studies were limited to flying insects and invertebrates on top of soils, and hence potential impacts of static electric fields on subsurface microorganisms and invertebrates went unnoticed. Nonetheless, power lines could influence soil and sediment communities and processes in at least two ways: First, shedding of ions provides a secondary source of pollution that may change direct current and ion transport in local environments with the consequences on microbial communities and processes described above. Second, strong variation in electric fields affect organisms using them for orientation (i.e., galvanotaxis or electrotaxis). Such behavioral responses have been observed for bacteria and invertebrates such as nematodes (Bespalov et al., 1996; Chrisman et al., 2016) and might further complicate the electrochemical environment in soils and sediments, which many organisms can alter (Traunspurger et al., 1997; Weerman et al., 2011; Hunting et al., 2013, 2015; Hunting and Kampfraath, 2013). Effects of local electrical pollution are readily amenable to tests in laboratory conditions by manipulating electric variables, but they can also be validated in natural settings (e.g., under power lines).

## CONCLUSION

Our results from experiments and field observations suggest that variation in AE can influence Earth's subsurface chemistry and the microorganisms in subsurface environments. We have provided proof of evidence that variations in AE can cascade down to changes in sediment redox conditions with implications for microbial electron transport activities and biogeochemical processes such as  $\text{SO}_4^-$  reduction. These insights widen our conceptual understanding of processes in water bodies, soils and sediments, and their overlooked links to AE. The coupling of AE

and subsurface electrochemistry is likely relevant to a wide range of organisms, in particular those with electrotactic behavior such as many microbial and nematode species. The proposed concept that AE could serve as a sinus node that sets the pace of Earth's biogeochemical heartbeat also presents many unknowns that call for pursuing diverse research avenues in the future.

## AUTHOR CONTRIBUTIONS

EH conceived, designed, and coordinated the study. RH and AK were involved in the initial conception. EH, AB, MV, and CC collected the field data. HvdG and PvB participated in the design of the study. HvdG, EH, and AK participated in the design of the conceptual figure. EH performed the experiments and statistical analyses. EH, RH, and MG drafted the manuscript. WA, PvB, AB, and HvdG contributed significantly to the earlier drafts. All authors contributed to improve the earlier drafts of the manuscript.

## ACKNOWLEDGMENTS

We thank Andrew Boulton, Martina Vijver, and Jack Middelburg for constructive comments on earlier drafts of the manuscript, Frank Hammecher, Gerard Muyzer, and Catarina Cuccio for useful discussions and assistance, and Hans Agema and Tijs van Roon for technical support. We are grateful to Dré Kampfraath and Frans Schupp for the graphical design of the conceptual diagram (Figure 1). We acknowledge the facilities provided by the Centro Biologia Alpina, Piora. Part of this work developed by participating in the COST Action 15211 Atmospheric Electricity Network: coupling with the Earth System, climate and biological systems, supported by the European Union COST (European Cooperation in Science and Technology) Program.

## SUPPLEMENTARY MATERIAL

The Supplementary Material for this article can be found online at: <https://www.frontiersin.org/articles/10.3389/fphys.2019.00378/full#supplementary-material>

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**Conflict of Interest Statement:** The authors declare that the research was conducted in the absence of any commercial or financial relationships that could be construed as a potential conflict of interest.

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