

# Soil pH governs production rate of calcium carbonate secreted by the earthworm Lumbricus terrestris

Article

Accepted Version

Lambkin, D. C., Gwilliam, K. H., Layton, C., Canti, M. G., Piearce, T. G. and Hodson, M. E. (2011) Soil pH governs production rate of calcium carbonate secreted by the earthworm Lumbricus terrestris. Applied Geochemistry, 26 (Supple). S64-S66. ISSN 0883-2927 doi: 10.1016/j.apgeochem.2011.03.032 Available at https://centaur.reading.ac.uk/20339/

It is advisable to refer to the publisher's version if you intend to cite from the work. See <u>Guidance on citing</u>.

To link to this article DOI: http://dx.doi.org/10.1016/j.apgeochem.2011.03.032

Publisher: Elsevier

All outputs in CentAUR are protected by Intellectual Property Rights law, including copyright law. Copyright and IPR is retained by the creators or other copyright holders. Terms and conditions for use of this material are defined in the <u>End User Agreement</u>.

www.reading.ac.uk/centaur



# CentAUR

Central Archive at the University of Reading

Reading's research outputs online

1	Soil pH governs production rate of calcium carbonate secreted by the earthworm Lumbricus
2	terrestris
3	
4	D.C. Lambkin <sup>a</sup> , K.H. Gwilliam <sup>a</sup> , C. Layton <sup>a</sup> , M.G. Canti <sup>b</sup> , T.G. Piearce <sup>c</sup> and M.E. Hodson <sup>a*</sup>
5	
6	<sup>a</sup> Soil Research Centre, School of Human and Environmental Sciences, University of
7	Reading, Whiteknights, Reading, RG6 6DW
8	<sup>b</sup> English Heritage, Ancient Monuments Laboratory, Centre for Archaeology, Fort
9	Cumberland, Eastney, PO4 9LD, UK
10	<sup>c</sup> Lancaster Environment Centre, Lancaster University, LA1 4YQ, UK
11	
12	* corresponding author. m.e.hodson@reading.ac.uk
13	

#### 1 ABSTRACT

*Lumbricus terrestris* earthworms exposed to 11 soils of contrasting properties produced, on average,  $0.8 \pm 0.1 \text{ mg}_{calcite}$  earthworm<sup>-1</sup> day<sup>-1</sup> in the form of granules up to 2 mm in diameter Production rate increased with soil pH (r<sup>2</sup> = 0.68, p ≤ 0.01). Earthworms could be a significant source of calcite in soils.

6

#### 7 1. INTRODUCTION

8 Earthworms secrete granules of calcium carbonate, predominantly calcite but also aragonite, 9 vaterite and amorphous calcium carbonate (Gago-Duport et al., 2008; Lee et al. 2008). The 10 granules are produced in the calciferous glands of the earthworm. In the case of L. terrestris 11 these occur in segments 10, 11 and 12 as three pairs of swellings off the oesophagus. 12 Micron-scale "spherites" of amorphous calcium carbonate are secreted in the rear two pairs of 13 oesophagal glands and move forwards into a pair of oesophagal pouches, by which time they 14 have largely crystallised to calcite and have combined to form granules. The granules are 15 secreted from the oesophagal pouches through a sphincter into the oesophagus. The function 16 that these granules serve for the earthworm is unknown (Darwin, 1881; Robertson, 1936; 17 Piearce, 1972; Briones et al., 2008) but previous studies have shown that earthworm granules 18 are commonplace in soils (Ponomareva, 1948; Wiecek and Messenger, 1972; Bal, 1977; 19 Canti, 1998). On the basis of field measurements Wiecek and Messenger (1972) estimated that excreted calcium carbonate could contribute up to 11 mol<sub>CaCO3</sub> ha<sup>-1</sup> yr<sup>-1</sup> to forest soils. 20 21 Canti and Piearce (2003) determined that granule production rates were greatest for the 22 earthworms Lumbricus terrestris and L. rubellus. Canti (2007) estimated production rates of 2.2 mg<sub>calcite</sub> day<sup>-1</sup> earthworm<sup>-1</sup> for *L. terrestris.* The aim of the study reported here was to 23 determine how granule production rate of *L. terrestris* varied with soil properties. 24

25

## 26 **2. METHODS**

Eleven soils were collected, air-dried, sieved to < 250 µm and characterised (Table 1). In all production experiments one (weighed) clitellate *L. terrestris* was added to moist soil (300 g air dry soil plus sufficient deionised water to raise the soil to 65 % of its water holding capacity).
Moisture content was kept constant throughout the experiment by addition of deionised water if treatments lost weight. Treatments were kept at 18 °C under ambient light conditions. At the
end of each experiment the earthworm was removed and weighed prior to release. The soil
was wet-sieved to 500 µm to recover freshly produced granules from the soil. These were air
dried and weighed.

5

6 In our first experiment earthworms were exposed to the 11 different soils for 27 days with six 7 replicates per treatment. Three grammes of horse manure were added to each earthworm 8 container at the start of the experiment. In a second experiment eight earthworms were each 9 separately and repeatedly exposed to Hamble soil for periods of 39 - 57 days. After each 10 exposure period granules were extracted from the soil, earthworms were weighed and then 11 transferred to fresh Hamble soil. This process was repeated a total of seven times over 315 12 days. Three grammes of horse manure were added to each container every 14 days. Data 13 were checked for normality and equality of variance. In the first experiment treatments were 14 compared using Kruskall-Wallis one-way analysis of variance on ranks and Dunn's method 15 for pairwise comparisons. In the second experiment reported correlations are Pearson 16 correlations. All statistical analysis used SigmaStat for Windows 3.01 produced by SPSS inc. 17 In the text values are expressed as mean + standard error.

18

## 19 3. RESULTS

20 At the end of the first experiment one earthworm had escaped or died from the Coombe 21 Complex, Kettering Loam, Neville, Park Gate and St Albans Wood soils and two from the 22 Hamble replicates. Granule production varied both within and between soils (p < 0.01) (Fig. 23 1). Within individual soils there was on average a factor of 7.7 ± 2.2 (n = 10, St Albans Wood 24 data discounted due to an absence of granules) times difference in masses of granules 25 produced by individual earthworms in the six replicate containers. Across all soils, granule 26 production rate varied between 0 in the pH 4.3 St Albans Wood soil (the next lowest production rate was 0.05 mg<sub>CaCO3</sub> earthworm<sup>-1</sup> day<sup>-1</sup> in one of the pH 6.1 Wilderness soil 27 replicates) and 4.3 mg<sub>calcite</sub> earthworm<sup>-1</sup> day<sup>-1</sup> (in one of the Coombe Complex soil replicates, 28 pH 7.8). Average production rate for all soils was  $0.8 \pm 0.1 \text{ mg}_{CaCO3}$  earthworm<sup>-1</sup> day<sup>-1</sup> (n = 29 30 57). Of the soil properties listed in Table 1, granule production rate was strongly correlated

- 1 with pH ( $r^2 = 0.68$ , p  $\leq 0.01$ ) but none of the other measured soil properties.
- 2

3	In the second experiment, over the 315 days two deaths occurred (between adjacent
4	sampling dates of 81 and 123 days) and the earthworms lost weight from an initial 5.3 $\pm$ 0.5 g
5	(n = 8,) to 2.4 $\pm$ 0.3 g (n = 6) at the end of the experiment. There was a significant correlation
6	between production rate and earthworm mass (r = 0.62, p $\leq$ 0.01). Three individual
7	earthworms (earthworms 1, 3 and 5), which showed a significant correlation between granule
8	production and earthworm mass (r = 0.65 to 0.92, p $\leq$ 0.01), also showed a significant
9	negative correlation between granule production and time (r = -0.78 to -0.93, p $\leq$ 0.01) (Fig.
10	2). On average granule production rate, expressed on a mg <sub>calcite</sub> earthworm <sup>-1</sup> day <sup>-1</sup> basis,
11	varied between earthworms by a factor of $3.4 \pm 0.6$ (n = 6); normalising to earthworm mass
12	(i.e. $mg_{calcite} g^{-1}_{earthworm} day^{-1}$ ) did not significantly reduce this level of variation (3.3 $\pm$ 0.5).
13	

#### 14 4. DISCUSSION

15 Granule production rate shows significant variation between earthworms and between soils. 16 The variation, at least in part, appears to reflect naturally occurring biological variation 17 between individual earthworms with larger earthworms producing more granules. However, 18 there are significant differences in production rates between soils, which indicates that some 19 of the variation is due to differences in soil properties. Whilst the correlation between 20 production rate and pH potentially reflects dissolution of the granules in the soil, granule 21 dissolution rates are not sufficiently rapid for this to be the case (Lambkin et al., this issue). 22 The hypothesis that the low soil pH lowers the saturation state of calcium carbonate in the 23 earthworm calciferous glands thereby limiting granule production is attractive. However, the 24 oesophagal glands where the granules are initially precipitated as spherites have no direct 25 contact with the soil and it seems unlikely that the fluids from which the spherites precipitate 26 would reflect the soil pH as they will have a homeostatically regulated pH. Set against this we 27 have observed inclusions of quartz and feldspar in granules (Lee et al. 2008) so there must 28 be some "leakage" of soil material into the calciferous glands, presumably this occurs as the sphincter opens and a granule is expelled from the glands into the oesophagus. Typically in 29 30 low pH soils there is less Ca (either total or available) and this could limit granule production.

However, in these experiments there was no discernable relationship between granule
 production and either total Ca levels in the soil or Ca present on exchange sites. Thus the
 cause of the relationship between soil pH and granule production remains unresolved.

4

An average value for earthworm density is 270 earthworms m<sup>-2</sup>, which covers a wide range of 5 species in temperature climates (Edwards and Bohlen, 1996); though not all earthworm 6 7 species produce granules. Using this value, granule production rates reported here of 0.05 to 0.8 to 4.3 mgCaCO<sub>3</sub> earthworm<sup>-1</sup> day<sup>-1</sup> (minimum, average and maximum values) correspond 8 to production rates of 493, 7 884 and 42 377 molCaCO<sub>3</sub> ha<sup>-1</sup> yr<sup>-1</sup>. A more conservative 9 estimate of 10 to 20 *L. terrestris* per m<sup>2</sup> (Briones et al. 2008) yields production rates of 18 to 10 3139 molCaCO<sub>3</sub> ha<sup>-1</sup> yr<sup>-1</sup> which are more similar to the estimates of Wiecek and Messenger 11 12 (1972) for forest soils. The precise amount of calcite produced annually by earthworms clearly 13 depends on estimates of earthworm numbers as well as soil properties; however, the 14 production rates of individual earthworms suggests to us that earthworm calcite production is 15 a potentially important source of calcite in soils.

16

#### 17 5. CONCLUSIONS

Biological variation has a significant impact on the mass of calcium carbonate secreted by earthworms in the form of granules; however, soil properties do play a role as well. The mass of granules produced by earthworms is significant and they are worthy of further study to improve our understanding of their significance in the terrestrial C cycle.

22

### 23 ACKNOWLEDGEMENTS

This work was funded by a NERC grant NE/F009623/1 awarded to MEH, TGP and MGC and also a summer studentship for CL funded by the Applied undergraduate research skills CETL hosted by the UoR. We thank Sir William Benyon for permission to collect soil on the Englefield Estate. As ever John Morgan and Pete Kille (Cardiff) provided inspiration through their enthusiasm and interest.

29

# 30 **REFERENCES CITED**

1	Bal, L., 1977. The formation of carbonate nodules and intercalary crystals in the soil by the
2	earthworm Lumbricus rubellus. Pedobiol. 17, 102–106.
3	Briones, M.J.I., Ostle, N.J., Piearce, T.G., 2008. Stable isotopes reveal that the calciferous
4	gland of earthworms is a $CO_2$ -fixing organ. Soil Biol. Biochem. 40, 554 – 557.
5	Canti, M.G., 1998, Origin of calcium carbonate granules found in buried soils and Quaternary
6	deposits: Boreas 27, p. 275–288.
7	Canti, M.G., 2007, Deposition and taphonomy of earthworm granules in relation to their
8	interpretative potential in Quaternary stratigraphy: J. Quarter. Sci. 22 111–118.
9	Canti, M.G. and Piearce, T.G., 2003, Morphology and dynamics of calcium carbonate
10	granules produced by different earthworm species: Pedobiol. 47, 511–521.
11	Darwin, C., 1881, The formation of vegetable mould, through the action of worms, with
12	observations on their habits. John Murray, London.
13	Edwards, C.A. and Bohlen, P.J., 1996, The biology and ecology of earthworms. Second ed.
14	Chapman and Hall, London.
15	Gago-Duport, L., Briones, M.J.I., Rodriguez, J.B. and Covelo, B., 2008, Amorphous calcium
16	carbonate biomineralisation in the earthworm's calciferous gland: Pathways to the
17	formation of crystalline phases. J. Struct. Biol. 162 422-435.
18	Hendershot, W.H. and Duquette, M., 1986, A simple barium chloride method for determining
19	cation exchange capacity and exchangeable cations. Soil Sci. Soc. Am. J. 50, 605–608.
20	Jarvis, R.A., 1968, Soils of the Reading District [Sheet 268]. Memoirs of the Soil Survey of
21	Great Britain, England and Wales: Harpenden, U.K., Rothamsted Experimental Station.
22	Kay, F.F., 1936, A soil survey of the University farm, Sonning, Berks: Reading, U.K.,
23	University of Reading, Faculty of Agriculture and Horticulture, Bulletin XLIX.
24	Lambkin, D.C., Canti, M.G., Piearce, T.G. and M.E. Hodson (this issue) Dissolution rates of
25	earthworm-secreted calcium carbonate. Appl. Geochem.
26	Lee, M.R., Hodson, M.E. and Langworthy, G., 2008, Earthworms produce granules of
27	intricately zoned calcite: Geology, 36, 943–946.
28	MAFF, 1986, Analysis of agricultural materials: A manual of the analytical methods used by
29	the Agricultural Development and Advisory service. Reference Book 427. Third ed.
30	London.

- 1 Piearce, T.G., 1972. The calcium relations of selected Lumbricidae: J.Animal Ecol. 41, 167–
- 2 **188**.
- Ponomareva, S.I., 1948, The rate of formation of calcite in the soil by the earthworms. Report
  of the Academy of Science of the USSR 61, 505–507.
- 5 Robertson, J.D., 1936, The function of the calciferous glands of earthworms: J. Expt. Biol. 13,
- 6 279–297.
- 7 Wiecek, C.S. and Messenger, A.S., 1972, Calcite contributions by earthworms to forest soils
- 8 in Northern Illinois. Soil Sci. Soc. Am. Proc. 36, 478–480.

Name <sup>1</sup>	Sample site	pН	LOI <sup>3</sup>	$WHC^4$	CEC⁵	Elemental composition <sup>6</sup> / wt %									Exchangeable ions <sup>7</sup> / mg kg <sup>-1</sup>						
						$AI_2O_3$	CaO	Fe <sub>2</sub> O <sub>3</sub>	K₂O	MgO	Na <sub>2</sub> O	$P_2O_5$	SiO <sub>2</sub>	AI	Ca	Fe	к	Mg	Na	Р	
Coombe Complex	SU625733	7.8 <u>+</u> 0.0	10.2 <u>+</u> 0.1	54.7 <u>+</u> 1.8	17.2 <u>+</u> 0.4	6	39	4	1	3	0	0	45	1.1 <u>+</u> 0.1	1933.6 <u>+</u> 133.6	0	88,2 <u>+</u> 5.9	59.3 <u>+</u> 2.4	12.0 <u>+</u> 2.8	36.6 <u>+</u> 0.2	
Frilsham	SU596726	7.0 <u>+</u> 0.0	7.7 <u>+</u> 0.1	48.7 <u>+</u> 1.9	16.6 <u>+</u> 0.2	8	1	3	2	0	0	0	84	1.6 <u>+</u> 0.0	3401.2 <u>+</u> 9.2	0	142.5 <u>+</u> 0.3	78.1 <u>+</u> 0.2	9.3 <u>+</u> 0.8	50.2 <u>+</u> 0.7	
Hamble	SU618702	7.9 <u>+</u> 0.0	4.2 <u>+</u> 0.0	40.8 <u>+</u> 0.8	10.7 <u>+</u> 0.1	8	1	3	2	1	0	0	84	1.3 <u>+</u> 0.0	2597.4 <u>+</u> 15.3	0	166.2 <u>+</u> 1.1	35.0 <u>+</u> 0.2	5.9 <u>+</u> 1.1	28.3 <u>+</u> 0.1	
Kettering	N/A <sup>2</sup>	7.4 <u>+</u> 0.0	8.8 <u>+</u> 0.1	53.8 <u>+</u> 1.0	24.5 <u>+</u> 0.5	14	2	8	2	1	0	0	71	2.1 <u>+</u> 0.1	4948 <u>+</u> 14	0.2 <u>+</u> 0.0	196.9 <u>+</u> 1.6	125.6 <u>+</u> 0.4	20.1 <u>+</u> 0.8	10.7 <u>+</u> 0.4	
Neville	SU765754	5.4 <u>+</u> 0.0	11.2 <u>+</u> 0.1	52.0 <u>+</u> 0.1	12.2 <u>+</u> 0.7	6	1	4	2	0	0	1	84	1.3 <u>+</u> 0.0	2533.7 <u>+</u> 5.4	0.2 <u>+</u> 0.0	473.2 <u>+</u> 3.4	262.4 <u>+</u> 0.4	13.3 <u>+</u> 1.6	102.9 <u>+</u> 1.0	
Parkgate	SU601684	5.6 <u>+</u> 0.0	8.5 <u>+</u> 0.1	61.0 <u>+</u> 2.8	15.4 <u>+</u> 0.3	10	0	4	2	1	0	0	81	1.3 <u>+</u> 0.0	2490.9 <u>+</u> 3.6	0.3 <u>+</u> 0.0	478.0 <u>+</u> 3.5	303.8 <u>+</u> 0.4	19.2 <u>+</u> 0.5	79.0 <u>+</u> 0.1	
Soil Science	SU731718	6.5 <u>+</u> 0.0	19.2 + 0.2	71.2 <u>+</u> 1.0	37.4 <u>+</u> 0.1	9	2	6	2	1	0	0	79	89.9 <u>+</u> 0.8	5515.8 <u>+</u> 17.2	88.8 <u>+</u> 1.7	455.2 <u>+</u> 11.7	312.7 + 1.5	307.8 + 1.9	20.4 + 0.6	
St Albans Field	SU603716	5.1 <u>+</u> 0.0	10.6 + 0.0	54.2 <u>+</u> 1.0	13.2 <u>+</u> 0.2	5	0	2	1	0	0	1	90	16.2 <u>+</u> 01	2402.2 <u>+</u> 3.9	3.9 <u>+</u> 0.1	262.1 <u>+</u> 1.8	72.4 + 0.1	6.7 <u>+</u> 0.2	82.3 + 0.8	
St Albans Wood	SU602716	4.3 <u>+</u> 0.0	90.2 + 1.0	143.2 + 2.8	35.1 <u>+</u> 0.1	10	4	4	2	0	0	1	77	2.2 <u>+</u> 0.0	5817.0 <u>+</u> 45.7	0.2 <u>+</u> 0.0	400.6 <u>+</u> 2.6	292.7 + 2.4	43.1 + 0.4	34.9 + 0.6	
Tidmarsh	SU626706	7.2 <u>+</u> 0.0	5.7 <u>+</u> 0.1	39.2 <u>+</u> 0.8	14.7 <u>+</u> 0.6	7	1	3	2	0	0	0	86	1.7 <u>+</u> 0.2	3324.0 <u>+</u> 487.7	0.1 <u>+</u> 0.0	244.7 <u>+</u> 33.6	57.8 + 8.5	12.8 + 1.4	44.0 + 0.1	
Wilderness	SU738715	6.1 <u>+</u> 0.0	27.6 <u>+</u> 0.1	78.1 <u>+</u> 1.4	42.4 <u>+</u> 0.1	10	3	5	2	1	0	1	79	2.7 <u>+</u> 0.1	6556.0 <u>+</u> 7.8	0.3 <u>+</u> 0.01	307.1 <u>+</u> 0.9	319.7 <u>+</u> 0.3	44.3 <u>+</u> 0.3	15.4 <u>+</u> 0.2	

Table 1. Mean chemical characteristics of the < 250 µm soils used in experiments expressed in terms of oven dry soil (n = 3 ± standard error except for elemental composition where n = 1)

<sup>1</sup>These correspond to the soil series from which the sample was taken (Jarvis, 1968; Kay, 1936) except for "Soil Science" and "Wilderness" which were sampled on the University of Reading campus; <sup>2</sup>Obtained commercially

from Broughton Loam and Turf Management, Kettering, UK; <sup>3</sup>Loss on ignition, %; <sup>4</sup>Water holding capacity, g<sub>H2O</sub> g<sup>-1</sup><sub>soil</sub>, <sup>5</sup>Cation exchange capacity, cmol<sub>c</sub> kg<sup>-1</sup> (Hendershot and Duquette, 1986); <sup>6</sup>By X-ray fluorescence,

normalised to 100 %; <sup>7</sup>Method of Hendershot and Duquette (1986) except P which is Olsen P (MAFF, 1986)



Fig. 1. Mean granule production rate for soils of different pH. Error bars = standard errors, n = 5 - 6.



Fig. 2. Granule production rate over time in the Hamble soil. After each sampling date earthworms were put into fresh soil. Earthworms 4 and 6 died at some point between the second (Day 81) and third (Day 123) sampling dates.