

Optimisation of the post-harvest conditions to produce chocolate aroma from jackfruit seeds

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1 ABSTRACT

Jackfruit seeds are an under-utilized waste in many tropical countries. In this work, 2 3 we demonstrate the potential of roasted jackfruit seeds to develop chocolate 4 aroma. Twenty-seven different roasted jackfruit seed flours were produced from 5 local jackfruit by acidifying or fermenting the seeds prior to drying, and roasting 6 under different time/temperature combinations. The chocolate aroma of groups of 7 four flours were ranked by a sensory panel (n=162) and response surface 8 methodology was used to identify optimum conditions. The results indicated a 9 significant and positive influence of fermentation and acidification on the 10 production of chocolate aroma. SPME/GC-MS of the flours showed that important 11 aroma compounds such as 2,3-diethyl-5-methylpyrazine and 2-phenylethyl acetate 12 were substantially higher in the fermented product, and that the more severe 13 roasting conditions produced 2-3 times more 2,3-diethyl-5-methylpyrazine, but less 14 3-methylbutanal. Moisture, a_w, pH, luminosity and color were also monitored to 15 ensure that these properties were similar to cocoa powder or cocoa substitutes.

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24 Keywords: jackfruit seeds, chocolate aroma, waste utilization, sensory analysis,
25 SPME/GC-MS

26 INTRODUCTION

Jackfruit (*Artocarpus heterophyllus* Lam.) is a large tropical fruit which is abundant in South America, Asia, Africa and Australia. It is a fleshy compound fruit (syncarp) belonging to the Moraceae family and takes 3-6 months to reach ripeness. The fruit weight ranges from 2 to 36 kg and its seeds account for around 15-18 % of the total weight of the fruit.^{1,2} Generally the seeds are boiled, steamed and roasted before eating, providing a cheap source of fiber, protein and minerals. In many countries, including Brazil, jackfruit seeds are an under-utilized waste stream.

There are several publications reporting the use of waste jackfruit seeds to 34 produce starch,³⁻⁶ but there is little reported in the literature on their potential to 35 36 generate flavor. For the first time we found that after roasting, jackfruit seeds imparted an aroma similar to chocolate. Chocolate aroma has been well-37 characterized^{7,8} and a number of different aroma compounds have been found to 38 39 contribute to the complex and characteristic aroma of chocolate. The most odor-active compounds in milk chocolate identified by Schnermann and Schieberle⁷ include 3-40 41 methylbutanal, phenylacetaldehyde and 2,3-diethyl-5-methylpyrazine and, a few more in roasted cocoa beans.⁹ Some pyrazines have been shown to contribute significantly 42 to the unique flavor of roast and toast foods⁹ and are used to determine the quantity 43 and quality of cocoa flavor.¹⁰ They impart chocolate, cocoa, hazelnut, roasted, coffee, 44 earth and green aromas.^{11,12} As with cocoa, the post-harvest pre-treatments and 45 46 roasting of the jackfruit seeds are likely to influence the formation of these compounds 47 and the quality of the aroma.

All three stages of the process (fermentation, drying and roasting) can have an influence on the final pyrazine concentration. During fermentation, enzymatic and microbial processes induce physical and chemical changes in seeds which result in

browning reactions.¹³ Some volatile compounds are formed at this stage, as well as 51 free amino acids and sugars which are substrates for the subsequent flavor-forming 52 reactions¹⁴ which take place during roasting. The influence of fermentation parameters 53 54 on the aroma of roasted cocoa beans is well understood and has been reviewed recently.¹⁵ Kirchhoff et al.¹⁵ demonstrated that chocolate aroma was correlated to 55 proteolysis and the subsequent accumulation of free amino acids. The proteolytic 56 57 enzymes such as endopeptidases and proteases are highly sensitive to pH, so pH control is important during cocoa fermentation to regulate the activity of different 58 enzymes. These products of fermentation (amino acids and reducing sugars) are the 59 precursors of pyrazines which are formed during roasting in the Maillard reaction.^{16–19} 60

Cocoa (*Theobroma cacao*) is a culture which is highly sensitive to changes in climate, is susceptible to many typical diseases and local farmers struggle to compete with international cocoa suppliers.^{20,21} Global cocoa production is around 3.7 million tons and this is not expected to grow significantly in the next 10 years,²² however demand by 2020 is estimated to be 4.5 million tons.²¹ In this context, new sources of chocolate aroma and flavor are important to meet the increase in demand and provide alternative revenue streams for local farmers and communities in Brazil.

The aim of this work is optimize the production of chocolate aroma from jackfruit seeds by treating them under conditions similar to those used in the cocoa process. Seeds will be acidified or fermented prior to drying, and roasted under different time/temperature combinations. Sensory ranking tests will be used to assess the chocolate aroma and key aroma compounds will be analysed by SPME/GC-MS.

74 MATERIALS AND METHODS

Chemicals. Standards of 3-methylbutanal, phenylacetaldehyde, 2-phenylethyl
acetate, 2-methylpyrazine, 2,5-dimethylpyrazine, 2,6-dimethylpyrazine, 2,3dimethylpyrazine, 2,3,5-trimethylpyrazine, 2,3,5,6-tetramethylpyrazine, 2,3-diethyl-5methylpyrazine, 1,2-dichlorobenzene and the alkane standards C₆–C₂₅ were purchased
from Sigma Aldrich Química, São Paulo, Brasil.

80 Jackfruit. Twenty five jackfruit of the hard pulp varieties were manually collected 81 from one single tree, between October 2013 and January 2014, in the countryside of 82 São Paulo, Brazil, selecting fruit of similar size (5 ± 1 kg) and ripeness, as indicated by 83 the yellow color of the shell. Jackfruits were cleaned manually in running water, the 84 seeds removed and the pulp discarded. These seeds were subjected to one of three 85 different treatments prior to roasting, producing either dried jackfruit seeds (DJS), acidified jackfruit seeds (AJS) or fermented jackfruit seeds (FJS). For each treatment, 86 87 the seeds from 7-9 jackfruit were pooled, and treated and dried in four x 1.5 kg 88 batches (3 kg batch for FJS) as described below. The dried beans (50 g from each of the 89 four batches) were roasted in 200 g portions. In total, 11 bags of roasted flour (200 g) 90 were prepared for each of the three treatments.

91 **Seed processing.** For the dried jackfruit seed (DJS), the seeds were dried in an 92 oven at 60 °C with air circulation. After 24 h, the spermoderms were manually 93 removed, and the seeds remained for a further 24 h in the same oven at the same 94 temperature.

For the acidified jackfruit seeds (AJS), treatment was carried out at ambient temperature (25 ± 3 °C). For each batch, the seeds (1.5 kg) were placed in polyethylene trays ($28 \times 42 \times 7.5$ cm) with 1% w/w acetic acid in potable water (3 kg). After five days the solution was removed and the seeds were dried using the same method as for DJS (2×24 h).

100 For the fermented jackfruit seeds (FJS), simulating what is done with cocoa, 101 seeds (3 kg) were placed in polyethylene boxes to ferment with added jackfruit pulp 102 (1.5 kg), perianth (0.52 kg), and banana leaves (0.1 kg) as a source of yeast. For the first 103 6-7 days of fermentation, the boxes were closed to promote anaerobic fermentation 104 but, for the remaining 7-8 days, the boxes were opened and the fermenting mass was 105 rolled daily to promote oxidation. The seeds were removed and dried using the same 106 method as for DJS (2 × 24 h). These processes are summarized in Figure 1A. The yield 107 from each treatment was expressed as in equation 1.

Yield (%) = (weight of flour after drying) x 100 / weight of raw jackfruit seeds 108 Eqn1 During acidification and fermentation, the ambient temperature and the 109 temperature of the fermenting mass were measured every 24 h²³ (AOAC Methods 110 111 13.010; 32.010; 32.016 and 32.017).. For AJS, the pH of an aliquot of liquid extracted 112 from the mass in the polyethylene boxes was measured directly. For the FJS mass, 10 g 113 of the fermenting mass was added to 100 mL of distilled waterIn both cases, the pH 114 was measured using a pH meter with a glass electrode standardized at the experiment 115 temperatures over the range from 7.0 to 4.0. For FJS, total titratable acidity (AOAC 945.08)²⁴ was measured using 5 g fermenting mass, diluted 10 times and filtered. 116 Proximate analysis was carried out according to Horwitz et al.²³ 117

Roasting and Grinding. For each treatment, 11 batches of seeds (200 g) were
roasted in a rotary electric oven (Probat[®] laboratory sample roaster, Emmerich am
Rhein, Germany) with digital temperature control, using conditions defined by the
response surface methodology. A central composite design was used for each
treatment (Figure 1B). Two factors (roasting time and temperature) were each tested
at five levels, with three repetitions of the central point totaling 11 samples. However,
preliminary experiments showed it was necessary to select different roasting

125 conditions for each treatment to avoid burning of the FJS yet achieve significant 126 roasting in the AJS. The temperature ranged from 150 to 201°C ± 0.1°C and the 127 roasting time from 33 to 47 min. The roasted seeds were then milled in a hammer mill 128 to produce a "flour". There was no heating of the sample during milling, minimizing 129 loss of volatile compounds at this stage. Flours were packed under vacuum and stored 130 without light at 8±1 °C.

131 Analysis of Flours. Water activity was determined from the temperature of the dew point (Aqualab[®]), moisture was determined by a standard gravimetric method, 132 and color was measured instrumentally using a Minolta[®] colorimeter, with illuminant C, 133 134 previously calibrated with a white surface (Y = 93.7, x = 0.3135 and y = 0.3195) based 135 on the CIE-lab L,* a,* b,* scale. The pH was determined in triplicate using 2 g of flour 136 added to distilled water (20 mL). The quality of the chocolate aroma was based on a 137 sensory comparison and the relative concentration of selected aroma compounds was 138 measured by GC-MS. The proximate composition was only carried out on flours with the highest sensory rankings. 139

Sensory analysis. All sensory evaluations were approved by the Ethics
Committee of Human Research of the ESALQ/USP (COET/077/131).

Preliminary sensory tests. Preliminary tests were carried out to determine the optimum temperature and time for sample exposure prior to the panelists receiving the sample for assessment. In this preliminary assay using AJS flour, the samples were placed in a water bath for five different combinations of time (30, 60, 120 s) and temperature (25, 36.5, 48 °C) prior to sniffing by a small panel comprising 21 untrained members aged 18-40 years (76% women). Each panelist was asked to rank groups of three samples in increasing order according to the intensity of the chocolate aroma

149 (Table 1). There was no significant difference between the conditions used to150 equilibrate the samples so the conditions were standardized at 40 °C for 120 s.

151 Sensorial ranking test. Ranking tests were used to determine the relative 152 intensity of chocolate aroma in 11 samples for each treatment (DJS, AJS and FJS) using 153 incomplete blocks (Figure 2). Each sample (3 g) was placed in an amber vial coded with 154 a random three-digit number and, prior to sensory evaluation, the vial was heated for 155 120 s in a water bath at 40 °C, these conditions having been selected from the 156 preliminary tests. Panelists received simultaneously four coded samples to rank in 157 increasing order of intensity of chocolate aroma, from least (=1) to most (=4). Total 158 ranking scores were used, thus the higher score representing the greater chocolate 159 aroma. The data obtained from the panelists were collected and analyzed using Compusense five." At the end of the session, the panelists were asked to describe 160 161 different aromas they identified in each of the samples using their own free choice of 162 descriptors.

Sensory experimental design. Untrained panelists (162) aged 18-54 years (60% 163 164 women) were randomly divided into three equal groups of 54 – one group for each 165 treatment (DJS, AJS and FJS). In order to reduce the number of comparisons to be 166 assessed by the panel, a balanced, incomplete block, design of experiment was used²⁵ 167 to construct a second order model based on the 11 samples in the central composite design.²⁵ However, to minimize panelist fatigue, the central point was represented by a 168 169 blend of the three central points (reducing the number of samples to 9) and the three 170 central points were assessed by the same panelists in a second sensory session.

171 In the first sensory session panelists received four samples in a balanced 172 incomplete block (Figure 2). Each sample block consisted of 18 comparisons (9 samples 173 each appearing 8 times). The sample block was repeated three times (for 54 panelists)

and the parameters, as defined by Cochran and Cox²⁴, were T=9; k=4; r=8; B=18; L=3;
E=84; Z=3.

176 In the second sensory session, the same panelists received three samples of the 177 central point (0, 0) (Figure 2). These samples were delivered at the same time, in a 178 randomized and balanced complete block²⁵. Total ranking scores from the second 179 session were transformed to be comparable to the first sensory session. Thus it was 180 possible to assess the variation between the central points and validate the response 181 surface for intensity of chocolate aroma.

Volatile analysis. Jackfruit flour (DJS, AJS and FJS) (3 g) was placed in a 20 mL SPME vial with 1 μ L of 1,2-dichlorobenzene in methanol (130.6 μ g/mL) and vortexed for 2 min. After equilibration at 45 °C for 15 min, the triple phase fiber (65 μ m PDMS/DVB/Carboxen from Supelco) was exposed (1cm) to the headspace above the sample for 55 min under magnetic agitation (635 rpm). These conditions had previously been optimized using surface response methodology.

188 The volatile compounds extracted by the fiber were analyzed by GC-MS using a 189 Shimadzu® QP2010 GC-MS equipped with a RTX5MS column (30 m, 0.25 mm i.d., 0.25 190 μ m film thickness). Volatile compounds were desorbed for 1 min during a splitless 191 injection at 200 °C. During desorption, the oven was maintained at 40 °C for a further 8 192 min, and then the temperature was raised at 4 °C/min to 200 °C, and then 10 °C/min to 193 280 °C totaling 56 min. MS was carried out using 70 eV electron impact, and m/z were 194 monitored in the range 40 to 500, in scan mode. Helium was the carrier gas and the flow rate was 1 mL/min in constant flow. A series of n-alkanes C₆-C₂₀ was analyzed 195 196 under the same conditions to obtain linear retention indices (LRIs) for comparison with 197 authentic samples. All volatile compounds listed were identified by comparison of their

198 mass spectrum and LRI with that of an authentic standard run under similar conditions.

199 Each sample was analyzed three times.

200 Peak areas for 3-methylbutanal, phenylacetaldehyde and 2-phenylethyl acetate 201 were measured using the total ion chromatogram. For the following compounds, the 202 peak area was approximated using the area of a characteristic m/z which was 203 multiplied by a factor calculated from the spectrum obtained from the authentic 204 standard: 2-methylpyrazine m/z 94, factor 3; 2,5/6-dimethylpyrazine (coelute) m/z205 108, factor 2.5; 2,3-dimethylpyrazine m/z 67, factor 4; trimethylpyrazine m/z 81, factor 206 14; tetramethylpyrazine m/z 54, factor 5; 2,3-diethyl-5-methyl-pyrazine m/z 121, 207 factor 20. The approximate relative concentration of each compound was obtained by 208 comparing the peak area against that of the internal standard (1,2-dichlorobenzene), 209 using 1 as a response factor.

210 Statistical analysis and response surface methodology. The central composite 211 design, Statistics[®] (2014), was selected for use. The two key responses were intensity of 212 chocolate aroma, 3-methylbutanal and 2,3-diethyl-5-methylpyrazine concentration, 213 although water activity, moisture, pH and color were also monitored. The two 214 independent variables of the design, roasting time and temperature, were coded as 215 x, and y, respectively. Equation 2 shows the quadratic polynomial model that was fitted to each response, where b_0 , b_1 , b_2 , b_{11} , b_{12} and b_{22} are the regression 216 217 coefficients; x and y are the values of the independent variables for roasting time (min) 218 and temperature (°C) respectively.

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$$z = b_0 + b_1 x + b_2 y + b_{11} x^2 + b_{22} y^2 + b_{12} x y$$
 Eqn 2

The analysis of variance (ANOVA) tables were generated and regression coefficients of individual linear, quadratic and interaction terms were determined by using design expert software (Statistics[®]). The significances ($p \le 0.05$) of all terms in the

223 polynomial model were judged statistically by computing the F value. XLStat was used 224 to carryout 2-way ANOVA on the volatile data and calculate Fisher's least significant 225 difference at p=0.05.

226 **RESULTS AND DISCUSION**

Jackfruit seed processing. In cocoa beans, control of the fermentation process is 227 required because unfermented beans develop little chocolate flavor, and excessive 228 fermentation may also result in unwanted flavors when roasted.²⁶ Generally for cocoa, 229 fermentation lasts between 5 to 8 days, and the end point is determined by 230 experience²⁷ based on reducing acid notes and maximizing chocolate flavor in the final 231 roasted product.²⁸⁻³⁰ 232

233 In this study, the fermentation process of 12 days was necessary, maybe because jackfruit seeds are bigger in comparison to cocoa beans and there is more substrate to 234 235 ferment. During the fermentation it is important to kill the embryo at the beginning to 236 ensure the success of the fermentation process and the formation of flavor compounds. Jinap and Dimick³¹ and Heemskerk et al.³² reported that a pH close to 4 237 would destroy the embryo; in jackfruit we found this value around day 3-4 of 238 239 fermentation (Figure 3A) whereas the acidification process started at pH 3 and 240 fluctuated between pH 3 and pH 4 (Figure 3B). In cocoa, samples are considered well fermented at pH > 4 although this varies with variety.²⁸ In practice, an increase in pH of 241 242 the seeds has been shown to improve chocolate flavor during fermentation and alkalization²⁹⁻³³ reported that pH values lower than 4.5 in the seeds decreased the 243 aromatic potential of the cocoa beans. So there is a balance between achieving a pH 244 245 which is low enough to kill the embryo but high enough to form aroma compounds.

Although titratable acidity in FJS was very variable, the overall trend was for an 246 increase as the pH dropped (Figure 3A and 3C). Rodriguez-Campos et al.³⁴ reported 247

similar results during cocoa fermentation with a correlation coefficient of -0.91
between pH and titratable acidity, and -0.86 for the correlation of the concentration of
acetic and lactic acid with pH.

251 Acetic and lactic acid are present in the first and second stages of cocoa 252 fermentation, when anaerobic yeasts and lactic acid bacteria are present, respectively. 253 Towards the end of fermentation, when aeration increases, the acetic acid bacteria become more significant. They are responsible for converting alcohol to acetic acid, 254 255 and since this reaction is exothermic (Figure 3D), it is likely that they are also responsible for the increase in temperature of the fermenting jackfruit mass.³² At the 256 257 end of the jackfruit fermentation period (day 12), the temperature of the mass had 258 risen from ambient to values near to 40 °C, similar to the rise during fermentation of cocoa beans although, in cocoa, temperatures can reach 45 °C.²⁸ Figure 3 shows the 259 260 pH, titratable acidity and temperature profile for FJS and pH for AJS.

For such a natural and variable process, these figures show that, with the exception of titratable acidity, these processes are fairly reproducible. In addition, it shows that jackfruit seeds can be fermented and dried under similar, albeit slightly longer, conditions to those applied to cocoa beans, resulting in a similar drop in pH which in cocoa results in the formation of aroma precursors.

Yield, pH, water activity (a_w), moisture, luminosity (L*) and chroma (c*) of jackfruit seed flours. In terms of total mass, the yields of flour obtained from DJS, AJS and FJS were 48%, 45% and 40% respectively.

The pH of the roasted jackfruit seed flours were highest (pH > 5) in the flours which had been roasted at the highest temperature (independent of seeds processing) and the lowest pH (< 4.9) was found in general in the FJS flour (Table 2). These pH values are similar to those reported in traditionally fermented and roasted cocoa (4.75

to 5.19).^{28,33-36} In other cocoa substitutes, Yousif and Alghzawi³⁷ found roasted carob
powder to be pH 4.81 and Queiroz and Garcia³⁸ reported the pH of roasted cupuaçu
flour as 4.77 - both similar to fermented and roasted jackfruit seeds (Table 2).

276 The pH of the flours can be fitted to a 3-dimensional surface as a function of time and temperature by using a combination of linear and quadratic terms, as well as an 277 interaction term, to construct a polynomial equation. The correlation coefficient (r^2) 278 279 indicates how well the data fit the model, and the p-value associated with each 280 coefficient in the equation indicates the certainty with which this term influences the response (Table S1). The correlation coefficient is good ($r^2 > 0.7$) so it is possible to 281 282 model and predict the pH of the flour from AJS and FJS as a function of time (x) and 283 temperature (y) using equations 3 and 4 respectively. In FJS flour (Eqn 4), there was a 284 linear and quadratic relationship with temperature (p = 0.006 and 0.03 respectively) 285 and a linear correlation with time (p = 0.04). For AJS flour we found significant linear effects with temperature (p =0.01). However, for DJS flour, the final pH was relatively 286 insensitive to changes in the roasting conditions and the model cannot be used 287 predictively (r^2 =0.6). The pH was on average higher in flours from DJS compared to AJS 288 289 and FJS.

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$$pH_{AJS}=37.455-0.42557x-0.27377y+0.001036x^{2}+0.000566y^{2}+0.0019444xy$$
 (r²=0.81)

291 Eqn 3

pH_{FJS}=19.94+0.0899x+0.2223y-0.00159x²+0.0006789y²+0.0002987xy (r²=0.93) Eqn 4
 Generally the moisture was associated with water activity (a_w) in flours, and both
 tended to decrease as roasting conditions became more severe (Table 2). In FJS flours,

the highest roast temperature (180 °C) for 40 min (0, 1.41) produced the lowest a_w and
the lowest moisture was obtained at 186-192 °C for a 35-40 min roast. In this study we
found 2.3% moisture in flour from FJS at (0, 1.41) which was high compared to DJS and

AJS flours roasted under similar conditions. By comparison, Yousif and Alghzawi³¹ 298 299 found 9.0 and 2.5% moisture for roast carob powder (150 °C for 60 min) and cocoa powder respectively, and Queiroz and Garcia³⁸ showed 3.0% moisture in roasted 300 301 cupuaçu powder. The a_w described for both these substitutes was around 0.4. Thus 302 flours of jackfruit seeds have similar or lower moisture and aw in comparison to cocoa 303 and other substitutes, which is important to restrict microbial growth in the flours and 304 for application in other products. The surface response design allows use of equations 305 5, 6 and 7 (x= time and y= temperature) to predict the moisture in the flour of DJS, AJS and FJS ($r^2 > 0.7$); in all equations we could observe the significant linear effect of both 306 307 roasting time and temperature in determining final moisture content (Table S1).

308 Moisture_{DJS}= $65.71-0.2845x-0.6427y-0.003497x^2+0.001512y^2-0.000383xy$ (r²=0.97;

309 linear temperature effect p=0.001) Eqn 5.

310 Moisture_{AJS}= $39.15-0.349499x-0.26448y-0.0067x^2+0.0006943y^2-0.0012498xy$ (r²=0.92;

311 linear temperature effect p=0.010) Eqn 6.

312 Moisture_{FJS}= $105.447-1.6209x-0.7213y+0.000926x^2+0.001246y^2-0.0050106xy$ (r²=0.90;

313 linear temperature effect p=0.020) Eqn 7.

In contrast, a_w, where there was much greater variability in the responses, can only be predicted in DJS flour and only the linear term in temperature was significant (equation 8), and negative, showing that as the temperature increased, the a_w decreased.

318 a_{wDJS}= -4.454 +0.1165x +0.03237y -0.0011974x² -0.000087113y²-0.000127xy
319 (r²=0.75; linear temperature effect p=0.04) Eqn 8.

Color in food is important because appearance can contribute to recognition, perception and enjoyment of the food. For substitutes, it is necessary to match the original product as closely as possible. In cocoa powder the luminosity (L*) is low (near 14

323 to black and brown), similar to the jackfruit flour which was produced from the high 324 temperature roasts. L* tended to be lower (darker) in FJS compared to AJS flour. For 325 chroma, the results were the reverse with high intensity color (larger chroma value) in 326 the higher roasts, and the FJS flours having the least intense color, although there 327 were few significant differences between roasting treatments. Luminosity results for fermented jackfruit seeds were similar to values in roasted cupuaçu. Cohen and Jackix, 328 ³⁹ reported L* of 42 in cupuaçu liquor compared to values of 50-70 found in the 329 jackfruit. Sacchetti et al.⁴⁰ found L* = 21 for roast cocoa beans (145 °C to 30 min) and 330 Sengül et al.⁴¹ found L*=19. Only Gu et al.¹² had slightly higher luminosity (L*= 41) for 331 roast cocoa (160 °C for 30 min). Therefore depending on the kind of product 332 333 developed using jackfruit seed flour, it may be necessary to modify the color with 334 other ingredients. It is possible to predict the luminosity and chroma of DJS flour using equations 9 and 10 (x= time and y= temperature, $r^2 > 0.7$). In both equations we 335 336 observed a significant negative linear effect ($p \le 0.05$) of roast temperature (i.e. as temperature increased, L* decreased and the product became darker), and, for DJS, 337 roast time was also significant. For acidified and fermented flours we found no 338 significant effect of roasting conditions (r² was 0.60 and 0.51 for chroma; and for 339 luminosity 0.44 and 0.52 for AJS and FJS respectively). 340

L*_{DJS}= 70.645 -0.9291x-0.2478y+0.005709x²+0.0000373y²+0.0060888xy (r²=0.94; linear
 temperature effect p=0.007; linear time effect p=0.0002) Eqn 9

343 Chroma_{DJS}=19.39 +0.27954x -0.47968y -0.008682 x^2 -0.00159939 y^2 + 0.002884xy344 (r^2 =0.88; linear temperature effect p=0.03) Eqn 10

345 Proximate composition of jackfruit seed flours. The proximate analysis was
346 only carried out on the three best roast conditions determined by sensory score (Table
347 3). For DJS flours, where there was no significant difference between the samples in
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348 terms of sensory score, a sample with high pyrazine content and a high sensory score 349 was selected. The different treatments produced different proximate composition. The 350 moisture was smallest in AJS flours, maybe because during five days in acetic acid 351 solution the seed had dehydrated. AJS and DJS were similar in proximate content. In 352 FJS, the fermentation process results in the breakdown of carbohydrates and the release of CO2. This is reflected in the proximate analysis where the remainder of the 353 354 material is assumed to be carbohydrate. This is significantly lower in FJS (53%) 355 compared to DJS (65%) and AJS (73%) respectively. The indirect consequence of this is a small increase in the % contribution from the other analytes. 356

Moisture, a_w, pH and color of the roasted jackfruit flours tended to vary with the time and temperature of the roasting conditions. However, the pH and moisture of the milled flours were similar to those of cocoa powder, and although the color was a bit pale (high L*), these flours have similar properties to cocoa, carob and cupuaçu powders, and could be used in similar products.

362 Sensory assessment of chocolate aroma of jackfruit flours. The response 363 surfaces for the sensory ranking tests are shown in Figure 4 A-C and the data are 364 shown in Table 4. The correlation coefficients for the 3D surface models for all three 365 processes (dry, acidified and fermented) were \geq 0.7. For DJS flours (Figure 4A), there 366 was no significant difference between samples ($p \le 0.05$) in the perception of sensory 367 chocolate aroma, although the model showed a linear effect with temperature ($p \le p$ 368 0.03) suggesting that the higher temperature may increase slightly the chocolate 369 aroma. For AJS flours, roasting at the temperature of the central point (180 °C) 370 generated the greatest sensory perception of chocolate aroma (Table 4). The model 371 showed a clear quadratic effect with temperature ($p \le 0.02$) shown in Figure 4B, which is also represented by a significant coefficient for y^2 ($r^2 = 0.86$) in the corresponding 372 16

373 equation, indicating a decrease in chocolate aroma as the roasting conditions became 374 more severe (and possibly over-cooked from a sensory perspective). However, the 375 most sensory chocolate aroma was found in FJS flours. The sensory rankings of 376 chocolate aroma (SCA) were 72 for FJS (40 min to 150 °C) compared to 70 for AJS (40 377 min to 180°C) and the average of DJS was 60. Clearly, a fermentation or acidification 378 process is necessary to produce chocolate aroma using jackfruit seeds, and it is 379 possible to select the best roasting conditions for each treatment to optimize the 380 sensory perception of chocolate aroma.

A range of descriptive terms were collected for the flours (Figure 5). All 381 382 treatments were described with chocolate and coffee terms. In addition, sweet aroma 383 attributes were used to described DJS flours (honey, milk, etc.) suggesting a relatively 384 mild processing treatment. Unfermented cocoa is very bitter and astringent with little apparent chocolate flavor ^{27,35} whereas the unfermented jackfruit flour (from DJS) still 385 386 had some chocolate aroma. For AJS flour, sweet aromas such as vanilla were similar to 387 DJS flour, but other descriptors were used (e.g. earthy, rancid, acid, silage, fermented, 388 green, etc.) which suggest that the chemical acidification process (rather than the 389 natural fermentation process) may produce less desirable attributes which are not 390 directly associate with food. However, FJS flour was described with fruity qualities 391 (orange, passion fruit, cherry, jackfruit and guava). These aromas are likely to be 392 related to fruity aldehydes, alcohols and esters which are products of the fermentation 393 process. FJS flour was also described with caramel, soya, hazelnut and roast attributes 394 suggesting a greater contribution from the Maillard reaction.

Overall, the sensory evaluation confirmed that a chocolate aroma can be generated from roasted jackfruit seeds, and demonstrated that it can be influenced by both the seed processing and the roasting conditions. The optimum chocolate aroma

score was obtained under moderate roasting conditions when the seeds had been fermented in a process similar to that used for fermenting cocoa beans, or acidified with acetic acid prior to roasting. However, the latter was described by the panel with additional less desirable terms. The best conditions were not necessarily obtained from the most severe roasting conditions and, for AFS flour, there was a very clear optimum, after which there was a decrease in chocolate aroma as the roasting conditions became more severe.

405 Volatile aroma compounds in jackfruit seed flours. Selection of aroma 406 compounds was based on a survey of the literature (1997-2017), considering only 407 those papers where the odor-active compounds in chocolate or other cocoa products had been established using GC-Olfactometry.^{7,8,25-27} From each paper, the 15-20 most 408 409 important aroma compounds for chocolate or cocoa aroma were identified and 410 collated, based on either their flavor dilution factors (FD), odor activity values (OAV) or 411 frequency of detection. The results of the survey are shown in Table S2. Chocolate 412 aroma is a complex mixture of 30-50 odor-active compounds, none of which imparts a 413 recognisable chocolate note. Some are present at very low concentrations (e.g. 2-414 acetyl-1-pyrroline, 4-hydroxy-2,5-dimethyl-3(2H)-furanone), often below the detection 415 threshold when using SPME. Others, although contributory, are reminiscent of aromas 416 very different to that of chocolate (e.g. 3-methylbutanoic acid, 2-methyl-3-417 methyldithio)furan and 1-octen-3-one which impart cheesy, meaty and mushroom 418 aromas respectively). In choosing just a few key compounds to monitor, our criteria 419 were based on selecting those which had previously been identified as having high FD 420 factors and high OAVs in chocolate or cocoa products, those which were relatively 421 abundant, and those which had a relevant aroma. On this basis we selected 3-422 methylbutanal, one of the most abundant compounds and also one which at the

appropriate dilution can be described as cocoa and malty. Phenylacetaldehyde and 2phenylethyl acetate were selected as compounds which contribute the floral character
to chocolate. 2,3-Diethyl-5-methylpyrazine and trimethylpyrazine were selected as
compounds which contribute the nutty earthy character. The approximate relative
contributions of these, plus four other pyrazines, are shown in Table 5.

The most obvious difference is the fact that in the FJS flours, all selected 428 429 volatiles, except 3-methylbutanal, were present at significantly higher concentrations 430 compared to the respective AJS and DJS flours, particularly the pyrazines, and 2phenylethyl acetate which was 50 times higher across all conditions. Since these 431 432 compounds are amongst those which have been shown most frequently to be 433 associated with chocolate aroma (Table S2), and have also been shown to be amongst 434 the most odor-active, it is highly likely that these compounds are responsible for the 435 high sensory scores for chocolate aroma in FJS.

Table 5 shows the significant differences within each pre-treatment (AJS, FJS or DJS). 2-Way ANOVA showed that for most compounds, under all treatments, there was a highly significant difference between flours prepared at different temperatures. In some cases, the roasting time was also significant, and the interaction between the two was significant in some cases.

It is interesting, however, that the key aroma compounds behaved quite differently with roasting time and temperature. With all three pre-treatments (AFS, FJS and DJS), 3-methylbutanal and phenylacetaldehyde showed a tendency to decrease as the more severe roasting conditions were employed. 3-Methylbutanal is both highly volatile and highly reactive: for example it readily undergoes aldol condensations with other aldehydes. Either or both of these may explain the decrease in concentration as the severity of the roasting process increased. This decrease in 3-methylbutanal may

also contribute to the decrease in chocolate aroma which was observed particularly in

449 AJS and also in FJS as the roasting conditions became more extreme.

450 The trends for 2-phenylethyl acetate were not clear or consistent, and within 451 each pre-treatment group, the differences due to different time-temperature 452 combinations were small or non-significant.

453 2,3-Diethyl-5-methylpyrazine, the most odor-active of the pyrazines identified in 454 most chocolate and cocoa products, showed a tendency to increase with increasing 455 severity of the roasting conditions, as is often the case for pyrazines. However, for 456 trimethylpyrazine, another important compound in chocolate aroma, the trends were 457 less clear, and in FJS it (and tetramethylpyrazine) tended to decrease with more severe 458 conditions, although both tended to increase slightly in AJS and DJS. The 459 dimethylpyrazines also tended to increase with increased roasting conditions in AJS 460 and DJS, but did not vary much in FJS.

In AJS and DJS, as the roasting conditions became more severe, the 3methylbutanal decreased whereas the 2,3-diethyl-5-methylpyrazine increased. Both being important for chocolate aroma, this is consistent with the sensory data which showed an optimum sensory chocolate aroma under moderate roasting conditions for AJS and DJS. In addition, the more severe conditions might also promote the formation of other pyrazines which at higher concentration would impart more roasted and burnt notes, as described in some DJS and FJS samples.

In FJS, most of the compounds were not sensitive to changes in roasting conditions, although 3-methylbutanal, phenylacetaldehyde and trimethylpyrazine tended to decrease. This is consistent with the sensory perception of chocolate aroma in FJS which showed a tendency to decrease as the roasting temperature increased.

472 Response surface methodology. The response surfaces for 2,3-diethyl-5-473 methylpyrazine are shown in Figures 6 A-C and the corresponding equations in Table 474 S1. The most noticeable difference between the treatments is the relative 475 concentration of 2,3-diethyl-5-methylpyrazine in FJS flour which was approximately 476 five and three times bigger than in flour from DJS and AJS respectively. Figure 6 clearly 477 demonstrates the positive influence of time and temperature on the formation of this 478 compound. However, in AJS and FJS flour, none of the coefficients relating to roast 479 time or temperature had a significant impact on the response at p<0.05, either linear 480 or quadratic, although they were significant at p<0.1. P-values were 0.07 and 0.09 respectively and positive, confirming the positive effect of temperature. 481

482 Direct comparison of the formation of 2,3-diethyl-5-methylpyrazine at the lowest 483 and highest temperature (t= 40 min in all cases) showed that it was significantly higher 484 in all three flours when the higher temperature was employed (Table 5). Furthermore, 485 in DJS, four out of the six pyrazines monitored also showed a significant increase (all at p<0.001) and in AJS five out of six showed a significant increase (four of these at 486 p<0.001). This is in agreement with many other studies^{10,37} that show that pyrazine 487 formation in general is greatly influenced by temperature. Queiroz and Garcia³² 488 489 evaluated roasting time and temperature for cupuaçu seeds and concluded that 490 increased time resulted in greater pyrazine formation and increased the scores for 491 chocolate in the sensory profile. For cupuaçu, the best roasting conditions were 150°C for 42 min. For cocoa beans, Farah et al.¹⁰ reported an increase in the concentration of 492 493 pyrazines, particularly tetramethylpyrazine, when they roasted beans at temperatures 494 close to 160°C.

495 Figure 6 (A, B and C) shows that the greatest relative concentration of 2,3-496 dimethyl-5-methylpyrazine was formed in dry, acidified and fermented flour when we

used 171 or 186, 201 and 180°C, respectively. These temperatures are higher than
those milder conditions (110 - 140 °C for 20 - 50 min) reported for cocoa by Jinap et
al.³⁶ or Afoakwa et al.¹³ (120-150 °C for 5-120 min).

500 The response surfaces for 3-methylbutanal are shown in Figures 6 D-F. They clearly demonstrate that, contrary to 2,3-diethyl-5-methylpyrazine, high time and 501 temperature are not the most favorable roasting conditions for the formation of 3-502 503 methylbutanal. The equation in Table S1 shows that the linear temperature coefficient 504 in AJS is significant (p=0.03) and negative, indicating that the lower temperatures produce a greater response. For AJS and FJS, the lowest temperatures generated the 505 506 most 3-methylbutanal, but in DJS, there was an optimum around the mid-point, 507 consistent with the data presented in Table 5. Optimum temperatures for 3-508 methylbutanal in DJS, AJS and FJS were 171, 165 and 154 °C, respectively, closer to 509 those used for cocoa roasting.

510 The similarity of the optimum jackfruit roasting conditions, compared to cocoa, 511 may be due to the fact that jackfruit seeds have a similar composition compared to 512 cocoa beans, although dried jackfruit seeds have a lower lipid content (0.4% compared 513 to dried cocoa beans which have range between 53 and 39%).^{12,13}

514 Whilst we have selected a few compounds as a marker of chocolate flavor, it is 515 clear from these results that there are other factors involved, particularly those 516 associated with the fermented product. Further work is currently being carried out to 517 investigate more thoroughly the contribution from a wider range of volatile 518 compounds.

519 Waste jackfruit seeds have been roasted to prepare a flour which has a chocolate 520 aroma. Moisture, pH and color were similar to those of cocoa, and different aroma 521 profiles were obtained by acidifying or fermenting the seeds prior to roasting under

- 522 different time/temperature combinations. Optimum chocolate aroma scores were
- 523 achieved when either fermentation or acidification was performed prior to roasting,
- and fermentation produced fewer off-notes. Utilization of this local waste stream can

525 provide a new revenue stream for local farmers and boost local economies.

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530 ASSOCIATED CONTENT

Table S1: Equations, coefficients, r² and p-value for all equations derived from the response surface methodology. Table S2 Summary of odor-active compounds found in chocolate and cocoa 1997-2017. This material is available free of charge via the Internet at http://pubs.acs.org.

535 REFERENCES

536 (1) Saxena, A.; Bawa, A. S.; Raju, P. S. Jackfruit (*Artocarpus heterophyllus* Lam.). In
537 *Postharvest Biology and Technology of Tropical And Subtropical Fruits*; Yahia, E.
538 M., Ed.; Woodhead: Philadelphia, 2011; pp 275–298.

- 539 (2) Mahanta, C. L.; Kalita, D. Processing and utilization of jackfruit seeds. In
 540 *Processing and Impact on Active Components in Food*; Preedy, V. R., Ed.;
 541 Elsevier: London, U.K., 2015; pp 395–400.
- 542 (3) Madruga, M. S.; De Albuquerque, F. S. M.; Silva, I. R. A.; Do Amaral, D. S.;
 543 Magnani, M.; Neto, V. Q. Chemical, morphological and functional properties of
 544 Brazilian jackfruit (*Artocarpus heterophyllus* L.) seeds starch. *Food Chem.* 2014,
 545 143, 440–445.
- 546 (4) Rengsutthi, K.; Charoenrein, S. Physico-chemical properties of jackfruit seed
 547 starch (*Artocarpus heterophyllus*) and its application as a thickener and stabilizer
 548 in chilli sauce. *Lebensm. Wiss. Technol.* 2011, 44, 1309–1313.
- 549 (5) Mukprasirt, A.; Sajjaanantakul, K. Physico-chemical properties of flour and
 550 starch from jackfruit seeds (*Artocarpus heterophyllus* Lam.) compared with
 551 modified starches. *Int. J. Food Sci. Technol.* 2004, *39*, 271–276.
- 552 (6) Tulyathan, V.; Tananuwong, K.; Songjinda, P.; Jaiboon, N. Some physicochemical
 553 properties of jackfruit (*Artocarpus heterophyllus* Lam) seed flour and starch.
 554 *ScienceAsia* 2002, 28, 37–41.
- 555 (7) Schnermann, P.; Schieberle, P. Evaluation of key odorants in milk chocolate and
 556 cocoa mass by aroma extract dilution analysis. *J. Agric. Food Chem.* 1997, 45,
 557 867–872.
- (8) Counet, C.; Callemien, D.; Ouwerx, C.; Collin, S. Use of Gas
 Chromatography–olfactometry to identify key odorant compounds in dark
 chocolate. Comparison of samples before and after conching. *J. Agric. Food Chem.* 2002, *50*, 2385–2391.
- 562 (9) Seitz, E. W. Fermentation production of pyrazines and terpenoids for flavour
 563 and fragrances. In *Bioprocess Production of Flavour, Fragrance, and Color*564 *Ingredients*; Gabelman, A., Ed.; Wiley: New York, 1994; pp 95–134.
- 565 (10) Farah, D. M. H.; Zaibunnisa, A. H.; Misnawi. Optimization of cocoa beans
 566 roasting process using response surface methodology based on concentration of
 567 pyrazine and acrylamide. *Int. Food Res. J.* 2012, *19*, 1355–1359.
- 568 (11) Tran, P. D.; Van de Walle, D.; De Clercq, N.; De Winne, A.; Kadow, D.; Lieberei,
 569 R.; Messens, K.; Tran, D. N.; Dewettinck, K.; Van Durme, J. Assessing cocoa
 570 aroma quality by multiple analytical approaches. *Food Res. Int.* 2015, *77*, 657–
 571 669.

572 (12) Gu, F.; Tan, L.; Wu, H.; Fang, Y.; Xu, F.; Chu, Z.; Wang, Q. Comparison of cocoa 573 beans from China, Indonesia and Papua New Guinea. Foods 2013, 2, 183–197. 574 (13) Afoakwa, E. O. Cocoa cultivation, bean composition and chocolate flavour 575 precursor formation. In Chocolate Science and Technology; Afoakwa, E. O., Ed.; 576 Wiley Blackwell: Oxford, U.K., 2010; pp 12–34. 577 (14)Parker, J. K. Thermal generation of aroma. In Flavour Development, Analysis and 578 Perception in Food and Beverages; Parker, J.K.; Elmore, J.S.; Methven, L. Eds.; 579 Woodhead Publishing: Cambridge, U.K., 2015; pp 3–30. 580 Kongor, J. E.; Hinneh, M.; de Walle, D. Van; Afoakwa, E. O.; Boeckx, P.; (15) 581 Dewettinck, K. Factors influencing quality variation in cocoa (*Theobroma cacao*) 582 bean flavour profile - a review. Food Res. Int. 2016, 82, 44–52. 583 (16)Kirchhoff, P. M.; Biehl, B.; Crone, G. Peculiarity of the accumulation of free 584 amino acids during cocoa fermentation. Food Chem. 1989, 31, 295-311. 585 Koehler, P. E.; Odell, G. V. Factors affecting the formation of pyrazine (17)586 compounds in sugar-amine reactions. J. Agric. Food Chem. 1970, 18, 895-898. 587 Ito, K.; Mori, M. Formation of pyrazines in aqueous maltose / glucose / fructose-(18) 588 glutamine model systems upon heating at below 100°C. Food Sci. Technol. Res. 589 **2004**, *10*, 199–204. 590 Ledl, F.; Schleicher, E. New aspects of the Maillard reaction in foods and in the (19)591 human Bbody. Angew. Chemie-International Ed. 1990, 29, 565–594. 592 Oyekale, a. S.; Bolaji, M. B.; Olowa, O. W. The effects of climate change on cocoa (20)593 production and vulnerability assessment in Nigeria. Agric. J. 2009, 4, 77-85. 594 (21) Fairtrade Foundation. Fairtrade and Cocoa - Commodity Briefing. 2011, No. 595 August, 1–17. 596 (22) FAO - Food and Agriculture. Perspectivas a plazo medio de los productos básicos 597 agrícolas. http://www.fao.org/docrep/007/y5143s/y5143s0w.htm (accessed Oct 598 6, 2015). 599 (23) AOAC Official Methods of Analysis. Official methods of analysis of the 600 association of official analytical chemists, second edition; Sidney Williams, Ed.; 601 Arlington: Virginia, 1984. 602 AOAC Official Methods of Analysis. Official methods of analysis of the (24) 603 association of official analytical chemists. In Official methods of analysis of the 604 association of official analytical chemists; Horwitz, D. W., Dr. George W. Latimer, 605 J., Eds.; 2010; p 6. 606 (25) Cochran, W. G.; Cox, G. M. Experimental Designs, 2nd ed.; Cochran, W. G., Cox, 607 G. M., Eds.; John Wiley and Sons Inc: New York, 1950.

- 608 (26) Dimick, P. S.; Hoskin, J. C. The chemistry of flavour development in chocolate. In
 609 *Industrial chocolate manufacture and use*; Beckett, S. T., Ed.; Oxford, U.K., 1999;
 610 pp 137–387.
- 611 (27) Beckett, S. T. *Industrial chocolate manufacture and use*, 3rd ed.; Beckett, S. T.,
 612 Ed.; Blackwell Science, Oxford, U.K., 1999.
- 613 (28) Jinap, S.; Dimick, P. S. Acidic characteristics of fermented and dried cocoa beans
 614 from different countries of origin. *J. Food Sci.* **1990**, *55*, 547.
- 615 (29) Heemskerk, R. F. M. Cleaning, roasting and winnowing. In *Industral chocolate*616 *manufacture and use*; Beckett, S. T., Ed.; Blackwell Science: Oxford, U.K., 1999;
 617 pp 78–100.
- 618 (30) Rodriguez-Campos, J.; Escalona-Buendía, H. B.; Orozco-Avila, I.; Lugo-Cervantes,
 619 E.; Jaramillo-Flores, M. E. Dynamics of volatile and non-volatile compounds in
 620 cocoa (*Theobroma cacao* L.) during fermentation and drying processes using
 621 principal components analysis. *Food Res. Int.* 2011, 44, 250–258.
- 622 (31) Yousif, A. K.; Alghzawi, H. M. Processing and characterization of carob powder.
 623 Food Chem. 2000, 69, 283–287.
- 624 (32) Queiroz, M. B.; Garcia, N. H. P. Evaluation to sensory profile to roast cupuassu
 625 and cocoa using QDA. 2000, 249–266.
- 626 (33) Cohen, K. D. O.; Jackix, M. D. N. H. Estudo do liquor de cupuaçu. *Ciência e* 627 *Tecnol. Aliment.* 2005, *25*, 182–190.
- 628 (34) Sacchetti, G.; Ioannone, F.; De Gregorio, M.; Di Mattia, C.; Serafini, M.;
 629 Mastrocola, D. Non enzymatic browning during cocoa roasting as affected by
 630 processing time and temperature. *J. Food Eng.* 2016, *169*, 44–52.
- (35) Hashim, P.; Selamat, J.; Syed Muhammad, S. K.; Ali, A. Changes in free amino
 acid, peptide-N, sugar and pyrazine concentration during cocoa fermentation. *J. Sci. Food Agric.* 1998, *78*, 535–542.
- (36) Jinap, S.; Wan Rosli, W. I.; Russly, a. R.; Nordin, L. M. Effect of roasting time and
 temperature on volatile component profiles during nib roasting of cocoa beans
 (*Theobroma cacao*). J. Sci. Food Agric. **1998**, 77, 441–448.
- 637 (37) Owusu, M.; Petersen, M.A.; Heimdal, H. Effect of fermentation method, roasting
 638 and conching conditions on the aroma volatiles of dark chocolate. J. Food
 639 Process. Pres.2012, 36, 446-456.
- 640 (38) Frauendorfer, F.; Schieberle, P. Identification of the key aroma compounds in
 641 cocoa powder based on molecular sensory correlations. *J. Agric. Food Chem.*642 2006, 54, 5521- 529.

644 FIGURE CAPTIONS

Figure 1A. Summary of jackfruit seed processing. DJS, AJS and FJS are dried, acidifiedand fermented jackfruit seeds respectively.

Figure 1B. Central composite design using two factors each at 5 levels; DJS, AJS and FJS
are dried, acidified and fermented jackfruit seeds respectively.

Figure 2. Experimental design used for sensory ranking test where T= number of samples; k= number of samples in each ranking test; r= number of times each sample was shown within each block; B= number of panelists in each block; L= number of times the samples were shown together; E = dependability of the analysis; Z= number times the block was repeated.²⁵

Figure 3. Variables followed during the processing of the seeds prior to roasting: A = pHduring fermentation process; B = pH during acidification process; C = total titratable acidity (g/100g) during fermentation process and D = temperature (°C) during fermentation process; close = anaerobic 6-7 days; open = aerobic 7-8 days.

Figure 4. Response surfaces for roasted jackfruit seeds. A, B and C = total sensory
chocolate aroma (SCA) ranking score for flour from DJS, AJS and FJS respectively.

660 Figure 5. Representation of aroma attributes used freely by the panelists to describe

661 the roasted flours from fermented, dried and acidified jackfruit seed

Figure 6. Response surfaces for roasted jackfruit seeds. A, B and C = 2,3-diethyl-5-

663 methylprazine for flour from DJS, AJS and FJS respectively. D, E and F = 3-

664 methylbutanal from DJS, AJS and FJS respectively.

Table 1. Results from preliminary ranking experiment using different pre-exposure conditions of the roasted flours prior to ranking

	total ranking score					
coded	values	actu	ial values	for sensory chocolate		
Х	У	time (s)	temperature (°C)	aroma ^a		
1	-1	120	25	13 a		
-1	-1	30	25	15 a		
-1	1	30	48	16 a		
1	1	120	48	20 a		
0	0	60	36.5	22 a		
0	0	60	36.5	21 a		
0	0	60	36.5	19 a		

T=7; k=3; r=3; b=21; L=1; E=78 where T = number of samples; k = number of samples in each ranking test; r = number of times each sample was shown within each block; B=

number of panelists in each block; L= number of times the samples were shown together; E = dependability of the analysis; Z= number times the block was repeated²⁴. ^avalues with the same letter are not significantly different at p<0.05 Table 2. Mean \pm standard error (n=3) pH, water activity, moisture L* and chroma* of

the roasted jackfruit seed flours showing mean values.

x (time)	y (temp)	рН	a _w	moisture %	L*	chroma
Flour fron	n dried jack	fruit seeds (DJS)				
0	1.41	5.4 ± 0.04^{a}	0.33 ± 0.02^{d}	1.0 ± 0.3^{d}	53 ± 3 ^e	33.6 ± 0.2^{a}
1	1	5.3 ± 0.01^{d}	0.32 ± 0.02^{d}	1.2 ± 0.4^{d}	55 ± 1 ^{de}	33.2 ± 0.2^{ab}
-1	1	5.3 ± 0.02 ^{bcd}	0.32 ± 0.01^{d}	1.3 ± 0.3^{d}	58 ± 3 ^{cde}	32.1 ± 0.6^{bc}
-1.41	0	5.3 ± 0.02 ^d	$0.37 \pm 0.01^{\circ}$	3.7 ± 0.8 ^b	59 ± 3 ^{bcde}	32.1 ± 0.2 ^{bc}
1.41	0	5.4 ± 0.02^{b}	0.31 ± 0.01^{d}	2.0 ± 0.2^{cd}	59± 0.4 ^{bcde}	33.3 ± 0.6 ^{ab}
0	0	5.2 ± 0.01^{e}	$0.38 \pm 0.01^{\circ}$	3.4 ± 0.6^{bc}	$60 \pm 2^{\text{abcde}}$	32.9 ± 0.4 ^{ab}
0	0	5.3 ± 0.01^{d}	0.42 ± 0.03^{abc}	2.9 ± 0.5 ^{bc}	60 ± 3^{abcd}	33.2 ± 0.3 ^{ab}
0	0	5.2 ± 0.01^{e}	0.43 ± 0.01^{a}	3.2 ± 0.3^{bc}	60 ± 2^{abcde}	32.5 ± 0.5^{abc}
1	-1	5.2 ± 0.01^{e}	0.44 ± 0.01^{a}	5.8 ± 0.9^{a}	64± 2 ^{abc}	$31.6 \pm 0.5^{\circ}$
-1	-1	5.4 ± 0.00^{b}	0.40 ± 0.02^{abc}	5.8 ± 0.8^{a}	66.0 ± 1.6^{a}	$31.3 \pm 0.6^{\circ}$
0	-1.41	5.3 ± 0.01^{bc}	0.40 ± 0.01^{bc}	6.4 ± 0.2^{a}	65.1 ± 1.5^{ab}	$31.2 \pm 0.5^{\circ}$
Flour fron	n acidified j	ackfruit seeds (AJ	S)			
0	1.41	5.6 ± 0.01 ^ª	0.5 ± 0.01^{abc}	0.5 ± 0.4^{d}	61.7 ± 0.9 ^{cd}	32.0 ± 0.8^{a}
1	1	5.6 ± 0.01^{a}	0.50 ± 0.02^{a}	0.9 ± 0.9 ^{cd}	60.5 ± 1.6^{d}	31.5 ± 0.8^{a}
-1	1	5.2 ± 0.01^{f}	$0.47 \pm 0.01^{\circ}$	1.6 ± 0.2^{bcd}	65.3 ± 0.9^{abcd}	30.2 ± 0.2^{ab}
1.41	0	$5.3 \pm 0.01^{\circ}$	0.47 ± 0.01^{bc}	1.9 ± 0.7 ^{abcd}	66.1 ± 0.8^{abc}	31.1 ± 0.2^{a}
1.41	0	5.2 ± 0.01^{de}	0.48 ± 0.01^{abc}	2.4 ± 0.6^{abc}	64.7 ± 2.2 ^{abcd}	30.5 ± 0.7^{ab}
0	0	5.3 ± 0.01^{cd}	0.50 ± 0.01^{ab}	1.9 ± 0.7^{bcd}	66.1 ± 1.4^{abc}	30.2 ± 0.5^{ab}
0	0	5.2 ± 0.01 ^{ef}	0.47 ± 0.01^{bc}	2.0 ± 0.5^{bcd}	65.3 ± 0.4^{abcd}	30.8 ± 0.5^{ab}
0	0	4.9 ± 0.02 ^h	0.47 ± 0.01^{bc}	1.4 ± 0.7^{bcd}	64.3 ± 3.0 ^{abcd}	31.1 ± 1.0^{a}
1	-1	5.0 ± 0.02 ^g	$0.46 \pm 0.01^{\circ}$	2.6 ± 0.8^{ab}	63.2 ± 2.9 ^{bcd}	31.2 ± 0.8^{a}
-1	-1	5.2 ± 0.03 ^{ef}	0.48 ± 0.01^{abc}	2.9 ± 0.2 ^{ab}	67.3 ± 1.9 ^{ab}	30.6 ± 0.8^{ab}
0	-1.41	5.3 ± 0.01^{b}	0.48 ± 0.01^{abc}	3.8 ± 0.2^{a}	69.3 ± 0.8^{a}	29.1 ± 0.2^{b}
Flour fron	n fermente	d jackfruit seeds (I	FJS)			
0	1.41	$5.1 \pm 0.00^{\circ}$, 0.34 ± 0.01 ^e	2.3 ± 0.5^{f}	49.4 ± 1.9^{ab}	27.6 ± 0.5^{ab}
1	1	5.0 ± 0.01^{b}	0.42 ± 0.01^{ab}	2.3 ± 0.1^{f}	50.1 ± 1.8^{ab}	27.8 ± 0.4^{ab}
-1	1	$4.8 \pm 0.01^{\circ}$	0.39 ± 0.01^{cd}	2.5 ± 0.2^{f}	53.5 ± 1.3 [°]	28.1 ± 0.3^{ab}
-1.41	0	4.7 ± 0.01 ^g	0.38 ± 0.01^{d}	4.1 ± 0.5^{bcd}	52.3 ± 3.5 [°]	28.9 ± 0.7 ^ª
1.41	0	4.8 ± 0.01d ^e	0.43 ± 0.01^{a}	3.6 ± 0.1^{cde}	53.7 ± 0.6^{a}	28.8 ± 0.4^{a}
0	0	4.8 ± 0.02^{ef}	0.39 ± 0.01 ^{cd}	3.3 ± 0.3^{def}	52.8 ± 1.9 ^ª	27.7 ± 0.5 ^{ab}
0	0	4.7 ± 0.00^{f}	0.40 ± 0.01^{bc}	3.0 ± 0.4^{ef}	44.5± 3.4 ^b	26.8 ± 0.8^{b}
0	0	4.8 ± 0.01^{d}	0.37 ± 0.01^{d}	3.9 ± 0.5 ^{cde}	49.5 ± 2.1^{ab}	28.4 ± 0.4^{a}
1	-1	4.7 ± 0.01 ^g	0.41 ± 0.01^{ab}	4.5 ± 0.1^{bc}	51.5 ± 1.1 ^ª	27.7 ± 0.3 ^{ab}
-1	-1	4.5 ± 0.02 ^h	0.41 ± 0.01^{ab}	5.7 ± 0.2^{a}	51.3 ± 2.9 ^ª	27.9 ± 0.4 ^{ab}
0	-1.41	4.8 ± 0.01^{de}	0.38 ± 0.01^{cd}	5.1 ± 0.4^{ab}	49.0 ± 2.9^{ab}	27.9 ± 0.4^{ab}

Within each column for each treatment, values with the same letter are not significantly

different from each other ($p \le 0.05$) using the Tukey test.

Table 3. Proximate composition (% ± standard error) of jackfruit flours roasted under

the best condition	۱S.
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flour ^a	moisture	lipids	proteins	ash	fiber		
					insoluble	soluble	
DJS	5.88 ± 0.9 a ^b	0.40 ± 0.05 a	11.20 ± 0.7 b	2.90 ± 0.02 b	10.34±0.05 b	3.88±0.010 a	
AJS	1.38 ±0.7 b	0.30 ± 0.05 b	11.16 ± 0.5 b	2.44 ± 0.12 c	9.29±0.01 c	2.68±0.003 b	
FJS	5.10 ± 0.4 a	0.50 ± 0.03 a	14.82 ± 0.5 a	4.70 ± 0.09 a	18.9±0.8 a	3.34±0.002 a	

^aDJS = dried jackfruit seeds (47min at 171 °C); AJS = acidified jackfruit seeds (40 min at

180 °C); FJS =fermented jackfruit seeds (40 min at 150 °C)

^bMean (n=3), within each column, values with the same letter are not significantly different from each other ($p \le 0.05$) using the Tukey test.

Table 4. Total sensory chocolate aroma (SCA) ranking score for flour from DJS, AJS and

FJS.

	codeo	d values	tota senso	ores for e aroma ^a	
	x (time)	y (temp)	DJS	AJS	FJS
incomplete blo	ock				
1	1	-1	66 a ^b	67 ab	68 ab
2	-1	-1	52 a	58 ab	66 ab
3	-1	1	58 a	63 ab	60 ab
4	1	1	66 a	51 ab	51 bc
5	0	1.41	51 a	42 c	44 c
6	0	-1.41	57 a	54 bc	72 a
7	1.41	0	62 a	69 a	61 ab
8	-1.41	0	56 a	66 ab	55 bc
blend	0	0	62 a	70 a	63 ab
complete bloc	k ^c				
9	0	0	64 k	68 k	71 k
10	0	0	62 k	68 k	65 k
11	0	0	60 k	74 k	53 k

^aDJS = dried jackfruit seeds; AJS = acidified jackfruit seeds; FJS - fermented jackfruit seeds.

^bWithin each column, means followed by the same letters are not significantly different ($p \le 0.05$).

^cValues are transformed for comparison with incomplete block.

														significance ^c		
LRI ^ª	compound ID ^b					roa	sting conditi	ons					т	t	T×t	
DRIED	DRIED JACKFRUIT (DJS)															
	roasting temp	150 °C	156 °C	156 °C	171 °C	171 °C	171 °C	171 °C	171 °C	186 °C	186 °C	192 °C				
	roasting time	40 min	35 min	45 min	33 min	40 min	40 min	40 min	47 min	35 min	45 min	40 min				
657	3-methylbutanal	6.0d ^d	7.1 bcd	8.1 abc	8.8 ab	8.1 abc	9.7 a	6.5 cd	5.4 d	7.2 bcd	5.6 d	3.4 e	***	*	ns	
827	2-methylpyrazine	4.3 fg	3.3 g	4.6 fg	6.6 efg	15 d	12 de	11 def	124 a	25 c	33 b	32 b	***	***	ns	
916	2,5/6-dimethylpyrazine	39 e	54 e	66 de	101 cd	121 bc	124 bc	96 cd	123 bc	167 a	185 a	155 ab	***	ns	ns	
922	2,3-dimethylpyrazine	12 g	11 g	72 b	19 fg	30 def	34 d	21 efg	33 de	49 c	63 b	110 a	***	***	***	
1008	2,3,5-trimethylpyrazine	25 abc	23 abc	11 c	19 bc	25 abc	26 abc	19 bc	44 ab	42 a	36 ab	35 ab	*	ns	ns	
1058	phenylacetaldehyde	11.2a	8.9ab	9.7a	8.1ab	6.6abc	7.8ab	10.6a	6.8abc	6.8abc	1.9c	3.4bc	***	ns	ns	
1091	2,3,5,6-tetramethyl-	110 cde	92 cde	68 e	69 e	88 e	139 bcd	110 cde	91 de	180 ab	210 a	140 bc	***	ns	ns	
1157	2,3-diethyl-5-methyl- pyrazine	1.5 fg	1.3 g	1.9 ef	1.9 ef	2.6 cd	3.2 b	2.2 de	4.3 a	3.9 a	4.3 a	3 bc	***	***	ns	
1263	2-phenylethyl acetate	0.1 ab	0.1 ab	0.08 bc	0.09 bc	0.08 c	0.07 c	0.08 bc	0.12 a	0.13 a	0.12 a	0.08 c	**	***	ns	
ACIDI	FIED JACKFRUIT (AJS) roasting temp	159 °C	165 °C	165 °C	180 °C	180 °C	180 °C	180 °C	180 °C	195 °C	195 °C	201 °C				

Table 5. Approximate relative concentrations of selected volatiles in roasted jackfruit seed flours

	roasting time	40 min	35 min	45 min	33 min	40 min	40 min	40 min	47 min	35 min	45 min	40 min			
657	3-methylbutanal	11.2 bc	13.2 a	12.2 ab	10.3 c	10.1 c	4.4 ef	7.5 d	7.8 d	5.2 e	4 f	2.3 g	***	*	ns
827	2-methylpyrazine	2.7 d	3.5 d	11.6 c	1.2 d	1.6 d	28 a	21 b	11 c	2.6 d	25 ab	29 a	**	**	ns
916	2,5/6-dimethylpyrazine	66 d	74 d	110 c	110 c	120 bc	146 a	137 ab	86 d	113 c	140 ab	123 bc	***	***	ns
922	2,3-dimethylpyrazine	37 g	126 a	58 f	59 f	68 ef	97 c	79 d	8.7 h	73 de	110 b	110 bc	***	***	***
1008	2,3,5-trimethylpyrazine	52 abc	65 a	28 cd	30 bcd	35 abcd	44 abc	43 abc	18 d	45 abc	59 abc	65 ab	ns	*	*
1058	phenylacetaldehyde	10.7bc	11.8ab	10.8b	9.7bcd	7.4cd	14.8a	7.1d	3.3e	3.4e	2.7e	0.9e	ns	**	ns
1091	2,3,5,6-tetramethyl-	28 cd	67 bc	38 bcd	44 bcd	59 bcd	57 bcd	79 bc	12 d	88 b	160 a	150 a	***	ns	**
1157	2,3-diethyl-5-methyl-	1.2 f	1.1 f	1.8 ef	2.1 de	2.8 d	4.6 c	4.2 c	1.8 ef	4.9 bc	7.8 a	5.4 b	***	* * *	**
1263	2-phenylethyl acetate	0.13 d	0.14 d	0.16cd	0.14 d	0.17bcd	0.15d	0.19abc	0.07e	0.20ab	0.21a	0.19abc	***	***	ns
FERM	ENTED JACKFRUIT SEE	DS (FJS)													
	roasting temp	150 °C	154 °C	154 °C	165 °C	165 °C	165 °C	165 °C	165 °C	176 °C	176 °C	180 °C			
	roasting time	40 min	35 min	45 min	33 min	40 min	40 min	40 min	47 min	35 min	45 min	40 min			
657	3-methylbutanal	7 abcde	11 a	9.8 ab	9 abc	8.4 abcd	5.1 def	6.8 bcde	4.3 ef	5.9 cdef	2.3 f	2.8 f	***	*	ns
827	2-methylpyrazine	100 abc	86 cd	123 ab	86 bcd	100 abc	53 d	130 a	129 a	123 a	113 abc	110 abc	ns	ns	ns
916	2,5/6-dimethylpyrazine	375 ab	280 bc	450 a	250 bc	290 bc	234 c	312 bc	266 bc	278 bc	215 c	240 c	*	ns	*
922	2,3-dimethylpyrazine	487 ab	410 bc	600 a	420 bc	400 bc	421 bc	513 ab	510 ab	511 ab	402 bc	320 c	*	ns	**
1008	2,3,5-trimethylpyrazine	560 abc	480 bc	690 a	380 bcd	370 cd	110 e	130 e	98 e	210 de	130 e	270 de	***	ns	ns

1058	phenylacetaldehyde	20b	27a	20b	15bc	11cde	10cde	13c	12cd	10cde	6.9de	6.0e	***	ns	ns
1091	2,3,5,6-tetramethyl- pyrazine	4330 ab	3840 bc	5300 a	4380 ab	4070 b	3840 bc	5320 a	4190 b	4200 b	2700 d	2830 cd	**	ns	**
1157	2,3-diethyl-5-methyl- pyrazine	7.5 f	11 ef	14 bcd	12 de	16 abc	13 cde	18 a	16 ab	16 ab	19 a	15abc	***	*	ns
1263	2-phenylethyl acetate	7.3 c	12 a	11 ab	7.6 c	7 cd	6.6 cd	9.5 b	10.1 b	7.6 c	6.3 cd	5.5 b	***	*	ns

^{*a*} Linear retention index on RTX5MS column (30m), calculated from a linear equation between each pair of straight chain alkanes C_6-C_{30} .

^bIdentity: identity of all compounds confirmed by comparison of mass spectrum and LRI with that of the authentic standard run under similar conditions.

^cS: Significance of differences between samples within one pre-treatment (AJS, FJS or DJS) - probability, obtained from 2-way ANOVA, that there is a difference between means; ns = no significant difference between means (p>0.05); * significant at the 5% level; ** significant at the 1% level; *** significant at the 0.1% level,

with respect to; T = roasting temperature, t = roasting time, T×t interaction between roasting time and temperature.

^d Mean (n=3) relative concentration (µg/kg) = peak area of compound × concentration of internal standard (ISTD) / peak area of ISTD, nd = not

detected. Within each row, cells containing the same letter are not significantly different from each other at p<0.05.





Figure 2



Figure 3







Figure 5



Figure 6

TOC GRAPHIC

