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**Antioxidant activity, total phenolics and flavonoids contents: should we
ban *in vitro* screening methods?**

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Abstract

As many studies are disclosing the association between the ingestion of bioactive compounds and a decreased risk of noncommunicable diseases, the scientific community has shown much interest in these compounds. In addition, as bioactive compounds are regarded as reducing agents, hydrogen donors, singlet oxygen quenchers or metal chelators, the measurement of antioxidant activity by *in vitro* assays has become very popular in the last decades. Measuring the levels of total phenolics, flavonoids, and other (sub)classes using spectrophotometry represents a chemical index but chromatographic techniques are necessary to establish structure-activity. For bioactive purposes, *in vivo* models are recommended or, at very least, different methods that employ distinct mechanisms of action need to be used. In this regard, some comments were made concerning the *in vitro* screening methods that will help one to design future research studies on “bioactive compounds”.

Keywords: Folin-Ciocalteu; antioxidants; bioavailability; colorimetric methods; functional properties; *in vivo* studies; HPLC.

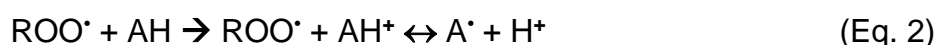
1. Phenolic compounds as antioxidants

Halliwell and Gutteridge (2007) state that “an *antioxidant* is a substance that, when present at a low concentration compared with that of an oxidizable substrate in the medium, inhibits oxidation of the substrate”. In this classification, phenolic compounds, which are derived from the secondary metabolism of plants, can protect multiple organs from oxidation. Therefore, phenolic compounds are regarded as natural *antioxidants*.

Antioxidants are categorized based on their *Function* (free-radical scavengers, scavengers of non-radical oxidizing agents, compounds that inhibit the generation of oxidants, transition metal chelating agents, and compounds that are able to stimulate the production of endogenous antioxidant compounds); *Polarity* (water-soluble and liposoluble); *Source*: (*exogenous* or *endogenous*); *Mechanism*: Antioxidants can neutralize the deleterious action of reactive species of cell membranes mainly by three mechanisms: hydrogen atom transfer (HAT), electron transfer (ET), and the ability to chelate transition metals (Prior et al., 2005; Brewer, 2011). In this sense, the HAT mechanism measures the ability of an antioxidant (AH) to quench free radicals (*i.e.*, peroxy radical - ROO[•]) by hydrogen donation stabilizing the peroxy radical by resonance according to the Equation (1):



The ET-based assays measure the ability of AH to transfer one electron to reduce free radicals, pro-oxidant metals and carbonyls, which are based on Equation (2) (Huang et al., 2005; Apak et al., 2013):



HAT assays include the oxygen radical absorbance capacity (ORAC), inhibition of lipoperoxidation, crocin bleaching assay, and β -carotene bleaching assay. Similarly, ET methods are composed of cupric-ion reducing antioxidant capacity (CUPRAC), Folin-Ciocalteu's phenol reagent reducing ability, scavenging effects in relation to 1,1-diphenyl-2-picrylhydrazyl (DPPH), and 2,2'-azino-bis(3-ethylbenzothiazoline-6-sulphonic acid (ABTS), among others (Shahidi & Zhong, 2015).

Some criticisms related to these *in vitro* chemical assays are based on the inexistence of such free radicals (DPPH/ABTS) in humans and the complexity of the mechanism of reaction. In addition, a high *in vitro* antioxidant activity cannot be translated into "treatment/cure" of illnesses. For instance, in the ferric reducing ability of plasma (FRAP) assay, as the reaction is performed at low pH values (3.6), much criticism is made on the translation of this method into *in vivo* effectiveness and, therefore, it can only be considered a screening method to have an idea of the antioxidant capacity of the sample (Schaich, Tian, & Xie, 2015). Undoubtedly, as these chemical assays are low-cost, easy to perform, do not require ultra-sensitive equipment, they are used to assess both isolated compounds and extracts from complex food matrices.

The antioxidant activity of phenolic compounds has been studied using a wide variety of methods, including *in vitro*, *ex vivo*, and *in vivo* protocols. Usually, authors find a high degree of correlation between *in vitro* antioxidant activity and the total phenolic content and/or individual phenolics (Rodrigo et al., 2005). However, the association between *in vitro* and *in vivo* antioxidant methods is still debatable and the opinion of experts in the field is divided into the usefulness of such *in vitro* methods.

2. Should we ban *in vitro* screening method to assess the antioxidant activity?

Several assays can be used to screen the *in vitro* antioxidant capacity of plant extracts, such as ferrous-ion chelating activity (Carter, 1971), copper chelating activity (Saiga, Tanabe, & Nishimura, 2003), lipid peroxidation inhibition assay (Daker et al., 2008), CUPRAC (Apak et al., 2008), deoxyribose assay (Chen, Zhang, & Xie, 2005), photoreduction of nitro blue tetrazolium assay (Chen, Zhang, & Xie, 2005), superoxide dismutase mimetic activity (Naithani, Nair, & Kakkar, 2006), total reducing capacity using a modified Folin-Ciocalteu assay (Berker et al., 2013), scavenging of hydrogen peroxide (Ruch, Cheng, & Klaunig, 1989), and cell-based *in vitro* antioxidant activity (Kellett, Greenspan, & Pegg, 2018). Excellent reviews on several chemical *in vitro* and cellular-based assays to assess the antioxidant activity can be found elsewhere (Alves et al., 2010; Niki, 2010; López-Alarcón & Denicól, 2013; Shahidi & Zhong, 2015). Without a doubt, the most frequently used methods rely on the use of DPPH, ABTS, FRAP, and ORAC assays (Halliwell, 2012; Schaich, Tian, & Xie, 2015).

These methods have many *pros* and *cons*, as any other analytical method, but when the antioxidant activity is evaluated, these methods have particularities in relation to the mechanism of action of the AH, the type of target (*i.e.*, H₂O₂ or DPPH radical), reactional pH, reaction time and temperature, and the use of a standard to build an analytical curve that is used to give a quantitative result in terms of antioxidant activity (Forman et al., 2014). Therefore, no single *in vitro* antioxidant activity assay will reflect the “total” antioxidant effect (Apak et al., 2013; Berker et al., 2013).

Recently, Harnly (2017) stated that studies regarding the measurement of *in vitro* antioxidant activity and total phenolic content using the Folin-Ciocalteu reagent is not appropriate. The reasons are:

1. There is currently no accepted standard mechanism or method to measure the antioxidant activity;
2. Only state-of-the-art techniques to identify antioxidants (*i.e.*, flavonoids) should be used in scientific research;
3. Results of a method *X* (*i.e.*, FRAP) are (usually) not comparable with data obtained using the method *Y* (*i.e.*, DPPH) or even between laboratories; and
4. *Antioxidant* is a marketing term of questionable health and analytical value as epidemiological studies are inconsistent.

In this regard, it is unquestionable that “state-of-the-art” techniques, such as liquid chromatography-mass spectroscopy (LC-MS), to identify and quantify phenolic compounds in foods, beverages, and herbal extracts have high accuracy and precision. However, screening spectrophotometric methods should also be used to characterize these materials and have an idea of the total content of phenolic compounds in the matrix (Granato, Santos, Maciel, & Nunes, 2016).

Halliwell (2012) stated that “the consumption of mega-doses of antioxidants (*i.e.*, pills) have also generally failed to prevent human disease, in part because they do not decrease oxidative damage *in vivo*”. Individuality (*i.e.*, genetics, gender, and body mass index) and life habits (*i.e.*, exercising, drugs/alcohol abuse, and smoking) also play an important role in the oxidative status of humans. Although some studies show discrepancies and inconsistencies to show a clear association between consumption of phenolic

compounds and increase of the antioxidant status in humans (Frankel & German, 2006; Saldanha et al., 2016), the search for antioxidants should continue and any allegation on functionality should be supported by preclinical, clinical, and epidemiological studies.

As well known, *in vitro* antioxidant methods and the estimation of total phenolic content using colorimetric assays can be used not only to have an idea of the beneficial effects of the food/extract. For quality control of natural products (Guo, Sun, Yu, & Qi, 2017; Lv, Zhang, Shi, & Lin, 2017), the antioxidant activity measured by *in vitro* methods are very useful as a fingerprint of reference materials that can be used for comparison purposes with commercial samples. Therefore, trends are generally very useful for comparative purposes of samples of the same material. In food technology, *in vitro* antioxidant assays together with the total phenolic content may be of importance to assess the best cutting styles of fruits (Li et al., 2017). These examples illustrate the usefulness of *in vitro* methodologies that can be applied in the routine quality control programs of food companies worldwide. Without a doubt, interferences in these nonselective methodologies exist and this fact is well demonstrated when comparing high-performance liquid chromatography (HPLC) results with total contents of phenolic compounds. Nevertheless, we need to have something in mind: one cannot rule out the usefulness of *in vitro* results despite their imperfect nature.

To date, Williams, Soencer, and Rice-Evans (2004) stated that “phenolic compounds may exert modulatory actions in cells through actions at protein kinase and lipid kinase signaling pathways. A clear understanding of the mechanisms of action of flavonoids, either as antioxidants or modulators of cell signaling, and the influence of their metabolism on these properties are key to

the evaluation of these potent biomolecules as anticancer agents, cardioprotectants, and inhibitors of neurodegeneration”. In addition, Alam, Bristi, & Rafiquzzaman (2013) stated that “antioxidants may be of great benefit in improving the quality of life by preventing or postponing the onset of non-communicable diseases”.

In recent studies, the antioxidant activity of bioactive compounds measured by *in vitro* and *in vivo* models are associated in a way that, depending on the biomarker used to assess the oxidative stress, interesting conclusions with practical applications arise (Macedo et al., 2013; Yan, Chen, & Zheng, 2017; Sun et al., 2017; Villa-Hernández et al., 2017; Aouachria et al., 2017; Naeimi & Alizadeh, 2017; Donado-Pestana et al., 2018). Obviously, there is a need to demonstrate the mechanistic approach behind the antioxidant activity of polyphenols *in vivo*. Animal models (*i.e.*, rat, mouse, rabbit, and dog) and human studies (*i.e.*, preclinical and randomized double-blind placebo-controlled clinical trials) are more appropriate but also more expensive, complex, and time-consuming compared to chemical and cellular-based methods (Thompson, Pederick, Singh, & Santhakumar, 2017). The assessment of *in vivo* antioxidant activity should include the measurement the activity of endogenous enzymes and antioxidant gene expression compared to a placebo, for instance. The bioaccessibility of phenolic compounds should also be studied in detail during and, principally, after the gastrointestinal digestion because the bioavailability of antioxidants, such as polyphenols, is generally very low. If these antioxidants could be absorbed, there is sometimes an insufficient concentration of the antioxidants in target tissues for the activity to be the prevalent protective mechanism (Huang et al., 2017).

Another point of consideration is as follows: what is measured in the food is not fully representative for what is active in humans. As well stressed by Espín, González-Sarrías, and Tomás-Barberán (2017) and Granado-Lorencio, Blanco-Navarro, Pérez-Sacristán, and Hernández-Álvarez (2017), “the type and quantity of the carotenoid/phenolic compounds metabolites produced in humans depend on the gut microbiota composition and function. The beneficial effect biological upon carotenoid/polyphenols intervention varies considerably and the chronic use of large doses may lead to saturation effects and the loss of linearity in the response. Therefore, the final health effects of dietary polyphenols/carotenoids depend on the gut microbiota composition”. As the microbiota of each individual is unique, we cannot assume “functionality” based only on *in vitro* tests.

3. Finals remarks and conclusions

As a conclusion of this viewpoint, although there will be divergent opinions in the scientific community based on thousands of studies available, we cannot close our eyes to dietary antioxidants and ignore some *in vitro* screening methods (*i.e.*, total phenolic/total flavonoids contents and antioxidant activity measurements) as low-cost, high-throughput tools to discover potential antioxidant sources for human consumption.

In a perspective, manuscripts on antioxidant properties based solely on colorimetric methods (including the Folin-Ciocalteu assay) will become unacceptable in *Food Chemistry* from now on. Authors are encouraged to assay bioactive compounds using chromatographic techniques (*i.e.*, HPLC/LC-MS) and, preferably, there must be some biological tests using cell lines or simulated digestion, or at the very least, measurement of bioactivity (*i.e.*, antioxidant effect)

using multiple assays that employ different mechanisms of action (*i.e.*, HAT, ET, and metal chelation property).

References

Alam, M. D., Bristi, N. J., & Rafiquzzman, M. (2013). Review on *in vivo* and *in vitro* methods evaluation of antioxidant activity. *Saudi Pharmaceutical Journal*, 21, 143-152.

Alves, C. Q., David, J. M., David, J. P., Bahia, M. V., & Aguiar, R. M. (2010). Methods for determination of *in vitro* antioxidant activity for extracts and organic compounds. *Química Nova*, 33, 2202-2210.

Aouachria, S., Boumerfeg, S., Benslama, A., Benbacha, F., Guemmez, T., Khennouf, S., Arrar, L., & Baghiani, A. (2017). Acute, sub-acute toxicity and antioxidant activities (*in vitro* and *in vivo*) of *Reichardia picroide* crude extract. *Journal of Ethnopharmacology*, 208, 105-116.

Apak, R., Gorinstein, S., Böhm, V., Schaich, K. M., Özyürek, M., & Güçlü, K. (2013). Methods of measurement and evaluation of natural antioxidant capacity/activity (IUPAC Technical Report). *Pure and Applied Chemistry*, 85(5), 957-998.

Apak, R., Guclu, K., Ozyurek, M., & Celik, S. E. (2008). Mechanism of antioxidant capacity assays and the CUPRAC (cupric ion reducing antioxidant capacity) assay. *Microchimica Acta*, 160(4), 413–419.

Berker, K. I., Olgun, F. A. O., Ozyurt, D., Demirata, B., & Apak, R. (2013). Modified Folin–Ciocalteu antioxidant capacity assay for measuring lipophilic antioxidants. *Journal of Agricultural and Food Chemistry*, 61, 4783-4791.

- 250 Brewer, M. S. (2011). Natural antioxidants: sources, compounds, mechanisms of
251 action, and potential applications. *Comprehensive Reviews in Food Science and*
252 *Food Safety*, 10, 221-247.
- 253 Carter, P. (1971). Spectrophotometric determination of serum iron at the
254 submicrogram level with a new reagent (ferrozine). *Analytical Biochemistry*,
255 40(2), 450-458.
- 256 Chen, H., Zhang, M., & Xie, B. (2005). Components and antioxidant activity of
257 polysaccharide conjugate from green tea. *Food Chemistry*, 90, 17-21.
- 258 Daker, M., Abdullah, N., Vikineswary, S., Goh, P. C., & Kuppusamy, U. R. (2008).
259 Antioxidant from maize and maize fermented by *Marasmiellus* sp. as stabiliser of
260 lipid-rich foods. *Food Chemistry*, 107, 1092-1098.
- 261 Donado-Pestana, C. M., dos Santos-Donado, P. R., Daza, L. D., Belchior, T.,
262 Festuccia, W. T., & Genovese, M. I. (2018). Cagaita fruit (*Eugenia dysenterica*
263 DC.) and obesity: Role of polyphenols on already established obesity. *Food*
264 *Research International*, 103, 40-47.
- 265 Espín, J. C., González-Sarriás, A., & Tomás-Barberán, F. A. (2017). The gut
266 microbiota: A key factor in the therapeutic effects of (poly)phenols. *Biochemical*
267 *Pharmacology*, 139, 82-93.
- 268 Forman, H. J., Davies, K. J. A., & Ursini, F. (2014). How do nutritional antioxidants
269 really work: Nucleophilic tone and para-hormesis versus free radical scavenging
270 *in vivo*. *Free Radical in Biology & Medicine*, 66, 24-35.
- 271 Frankel, E. N., & German, J. B. (2006). Antioxidants in foods and health:
272 problems and fallacies in the field. *Journal of the Science of Food and Agriculture*,
273 86, 1999–2001.

- 274 Granado-Lorencio, F., Blanco-Navarro, I., Pérez-Sacristán, B., & Hernández-
275 Álvarez, E. (2017). Biomarkers of carotenoid bioavailability. *Food Research*
276 *International*, 99(2), 902-916.
- 277 Granato, D., Santos, J. S., Maciel, L. G., & Nunes, D. S. (2016). Chemical
278 perspective and criticism on selected analytical methods used to estimate the
279 total content of phenolic compounds in food matrices. *Trends in Analytical*
280 *Chemistry*, 80, 266-279.
- 281 Guo, Y., Sun, L., Yu, B., & Qi, J. (2017). An integrated antioxidant activity
282 fingerprint for commercial teas based on their capacities to scavenge reactive
283 oxygen species. *Food Chemistry*, 237, 645-653.
- 284 Halliwell, B. (2012). Free radicals and antioxidants: updating a personal view.
285 *Nutrition Review*, 70(5), 257-265.
- 286 Halliwell, B. and Gutteridge, J. M. C. (2007). Free Radicals in Biology and
287 Medicine. 4th edition. Oxford: Oxford University Press, 888p.
- 288 Harnly, J. (2017). Antioxidant methods. *Journal of Food Composition and*
289 *Analysis*, in press.
- 290 Huang, D., Ou, B. and Prior, R. L. (2005). The chemistry behind antioxidant
291 capacity assay. *Journal of Agricultural and Food Chemistry*, 53, 1841–1856.
- 292 Huang, S., Ma, Y., Zhang, C., Cai, S., & Pang, M. (2017). Bioaccessibility and
293 antioxidant activity of phenolics in native and fermented *Prinsepia utilis* Royle
294 seed during a simulated gastrointestinal digestion *in vitro*. *Journal of Functional*
295 *Foods*, 37, 354-362.
- 296 Kellett, M. E., Greenspan, P., & Pegg, R. B. (2018). Modification of the Cellular
297 Antioxidant Activity (CAA) assay to study phenolic antioxidants in a Caco-2 cell
298 line. *Food Chemistry*, In press

- 299 Li, X., Long, Q., Gao, F., Han, C., Jin, P., & Zheng, Y. (2017). Effect of cutting
300 styles on quality and antioxidant activity in fresh-cut pitaya fruit. *Postharvest*
301 *Biology and Technology*, 124, 1-7.
- 302 López-Alarcón, C., & Denicol, A. (2013). Evaluating the antioxidant capacity of
303 natural products. A review on chemical and cellular-based assays. *Analytica*
304 *Chimica Acta*, 763, 1-10.
- 305 Lv, H., Zhang, Y., Shi, J., & Lin, Z. (2017). Phytochemical profiles and antioxidant
306 activities of Chinese dark teas obtained by different processing technologies.
307 *Food Research International*, 100, 486-493.
- 308 Macedo, L. F. L., Rogero, M. M., Guimarães, J. P., Granato, D., Lobato, L. P., &
309 Castro, I. A. (2013). Effect of red wines with different *in vitro* antioxidant activity
310 on oxidative stress of high-fat diet rats. *Food Chemistry*, 137, 122-129.
- 311 Naeimi, A. F., & Alizadeh, A. (2017). Antioxidant properties of the flavonoid fisetin:
312 An updated review of *in vivo* and *in vitro* studies. *Trends in Food Science and*
313 *Technology*, 70, 34-44.
- 314 Naithani, V., Nair, S. & Kakkar, P. (2006). Decline in antioxidant capacity of Indian
315 herbal teas during storage and its relation to phenolic content. *Food Research*
316 *International*, 39, 176-181.
- 317 Niki, E. (2010). Assessment of antioxidant capacity *in vitro* and *in vivo*. *Free*
318 *Radical Biology and Medicine*, 49(4), 503-215.
- 319 Prior, R. L., Wu, X. & Schaich, K. (2005). Standard methods for the determination
320 of antioxidant capacity and phenolics in foods and dietary supplements. *Journal*
321 *of Agricultural and Food Chemistry*, 53, 4290–4302.

- 322 Rodrigo, R., Castillo, R., Carrasco, R., Huerta, P., & Moreno, M. (2005).
 323 Diminution of tissue lipid peroxidation in rats is related to the *in vitro* antioxidant
 324 capacity of wine. *Life Sciences*, 76, 889-900.
- 325 Ruch, R. J., Cheng, S. J., & Klaunig, J. E. (1989). Prevention of cytotoxicity and
 326 inhibition of intracellular communication by antioxidant catechins isolated from
 327 Chinese green tea. *Carcinogenesis*, 10, 1003–1008.
- 328 Saiga, A. I., Tanabe, S., & Nishimura, T. (2003). Antioxidant activity of peptides
 329 obtained from porcine myofibrillar proteins by protease treatment. *Journal of*
 330 *Agricultural and Food Chemistry*, 51(12), 3661-3667.
- 331 Saldanha, J. F., Leal, V. O., Rizzetto, F., Grimmer, G. H., Ribeiro-Alves, M.,
 332 Delaprene, J. B., Carraro-Eduardo, J. C. & Mafra, D. (2016). Effects of resveratrol
 333 supplementation in Nrf2 and Nf-kb expressions in nondialyzed chronic kidney
 334 disease patients: a randomized, double-blind, placebo-controlled, crossover
 335 clinical trial. *Journal of Renal Nutrition*, 26(6), 401-406
- 336 Schaich, K. M., Tian, X., & Xie, J. (2015). Hurdles and pitfalls in measuring
 337 antioxidant efficacy: A critical evaluation of ABTS, DPPH, and ORAC assays.
 338 *Journal of Functional Foods*, 14, 111–125.
- 339 Shahidi, F., & Zhong, Y. (2015). Measurement of antioxidant activity. *Journal of*
 340 *Functional Foods*, 18, 757-781.
- 341 Shen, Y., Zhang, H., Cheng, L., Wang, L., Qian, H., & Qi, X. (2016). *In vitro* and
 342 *in vivo* antioxidant activity of polyphenols extracted from black highland barley.
 343 *Food Chemistry*, 194, 1003-1012.
- 344 Sun, Y., Tao, X., Men, X., Xu, Z., & Wang, T. (2017). *In vitro* and *in vivo*
 345 antioxidant activities of three major polyphenolic compounds in pomegranate

346 peel: Ellagic acid, punicalin, and punicalagin. *Journal of Integrative Agriculture*,
347 16(8), 1808-1818.

348 Thompson, K., Pederick, W., Singh, I., & Santhakumar, A. B. (2017). Anthocyanin
349 supplementation in alleviating thrombogenesis in overweight and obese
350 population: A randomized, double-blind, placebo-controlled study. *Journal of*
351 *Functional Foods*, 32, 131-138.

352 Villa-Hernández, J. M., Mendoza-Cardoso, G., Mendoza-Espinoza, J. A., Vela-
353 Hinojosa, C., León-Sánchez, F. D., Rivera-Cabrera, F., Alia-Tejacal, I., & Pérez-
354 Flores, L. J. Antioxidant capacity *in vitro* and *in vivo* of various ecotypes of
355 Mexican plum (*Spondias purpurea* L.). *Journal of Food Science*, in press.

356 Williams, R. J., Soencer, J. P. E., & Rice-Evans, C. (2004). Flavonoids:
357 antioxidants or signaling molecules. *Free Radical Biology & Medicine*, 36, 838–
358 849.

359 Yan, F., Chen, X., & Zheng, X. (2017). Protective effect of mulberry fruit
360 anthocyanin on human hepatocyte cells (LO2) and *Caenorhabditis elegans* under
361 hyperglycemic conditions. *Food Research International*, 102, 213-224.